

A simple and effective synthesis of 3- and 4-((phenylcarbamoyl)oxy)benzoic acids

Urban Košak^a and Stanislav Gobec^{a,*}

^aFaculty of Pharmacy, University of Ljubljana, Aškerčeva 7, 1000 Ljubljana, Slovenia

*Corresponding author. Tel.: + 386 (1)4769585; fax: + 386 (1)4258031; e-mail: stanislav.gobec@ffa.uni-lj.si

Abstract

Phenserine, posiphen, tolserine and cymserine and its derivatives are experimental Alzheimer's disease drugs that contain a phenyl phenylcarbamate moiety that is responsible for their anti-Alzheimer activities. We have developed a simple (3 steps) and effective (overall yields 76–90%) method for preparing 3- and 4-((phenylcarbamoyl)oxy)benzoic acids which can be reacted with amines to produce phenyl phenylcarbamate moiety containing amides as new potential anti-Alzheimer disease drugs. The synthesized carboxylic acids are thus important building blocks with potential use in medicinal chemistry and drug discovery.

Keywords

((Phenylcarbamoyl)oxy)benzoic acids, phenyl isocyanates, carbamates, building blocks, Alzheimer's disease.

1. Introduction

Alzheimer's disease (AD) is a progressive neurodegenerative brain disorder.¹ The synaptic dysfunction and neurodegeneration in AD most severely affects the cholinergic system.² This decreases the levels of the neurotransmitter acetylcholine (ACh),³ which then

produces cognitive impairment and memory loss,⁴ characteristic for patients with AD. Several compounds are currently being evaluated in preclinical and clinical trials for efficacy in AD, including cholinesterase (ChE) inhibitors which increase the levels of ACh in the brain: phenserine, posiphen, tolserine and cymserine and its derivatives (Figure 1).⁵ These experimental Alzheimer's disease drugs all contain the phenyl phenylcarbamate moiety or its derivative. Phenserine⁶ and posiphen⁷ contain a phenyl phenylcarbamate moiety, tolserine⁸ contains a phenyl *o*-tolylcarbamate moiety and cymserine and its derivatives^{9,10} contain a phenyl (4-isopropylphenyl)carbamate moiety (Figure 1).

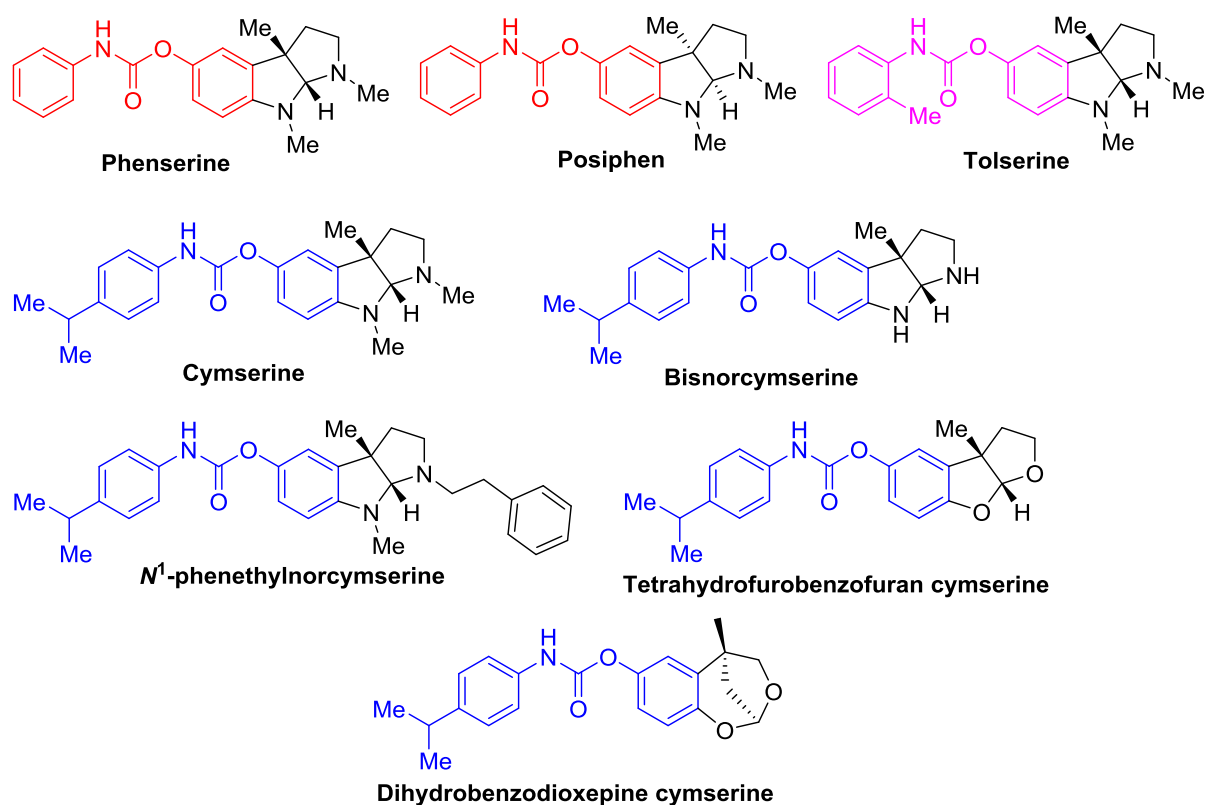
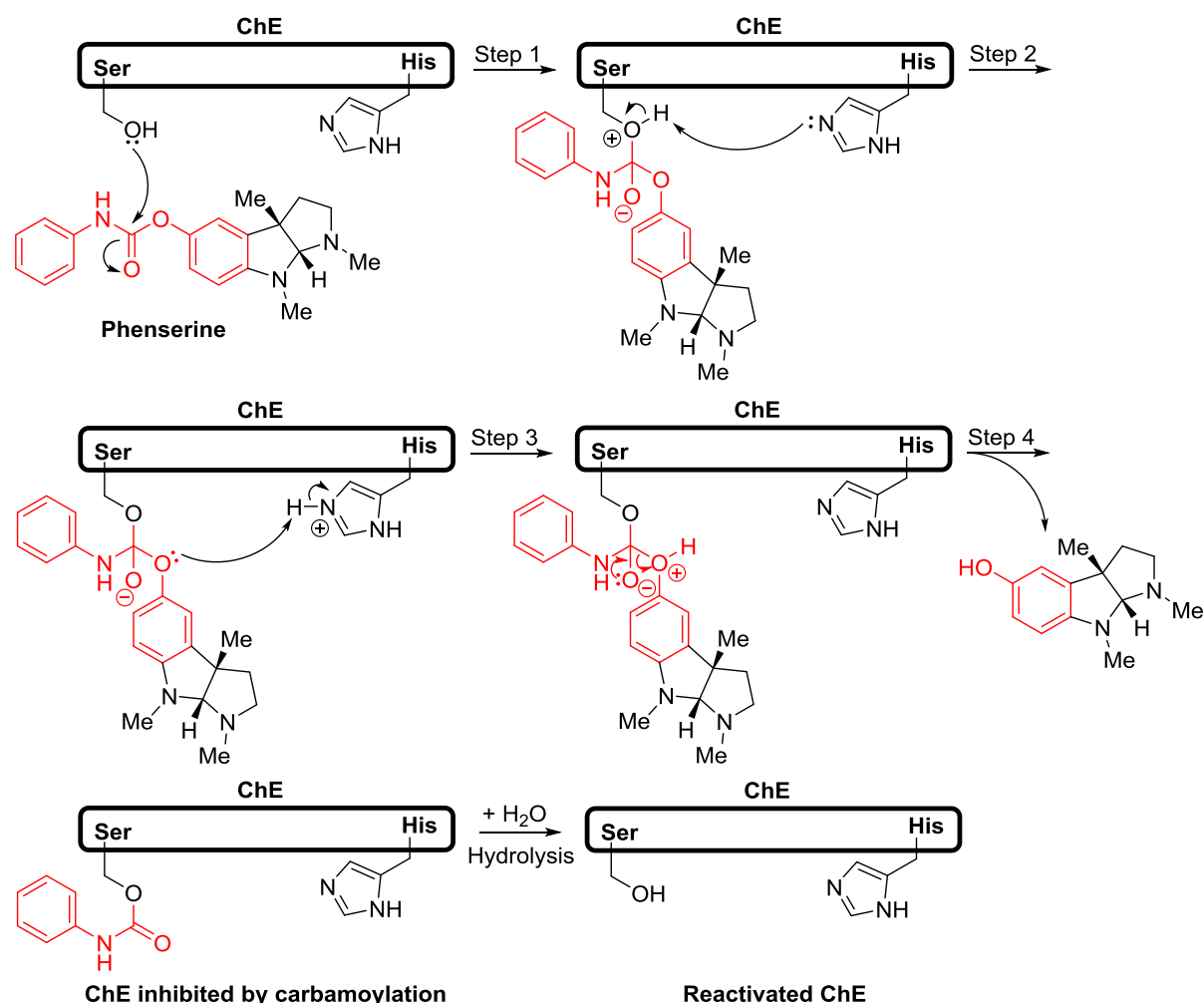


Figure 1. Structures of phenyl phenylcarbamate containing experimental Alzheimer's disease drugs.

Phenserine, posiphen, tolserine and cymserine and its derivatives are pseudo-irreversible carbamate inhibitors of ChEs where the phenyl phenylcarbamate moiety is responsible for their biological activity. Their mechanism of inhibition involves a rapid initial

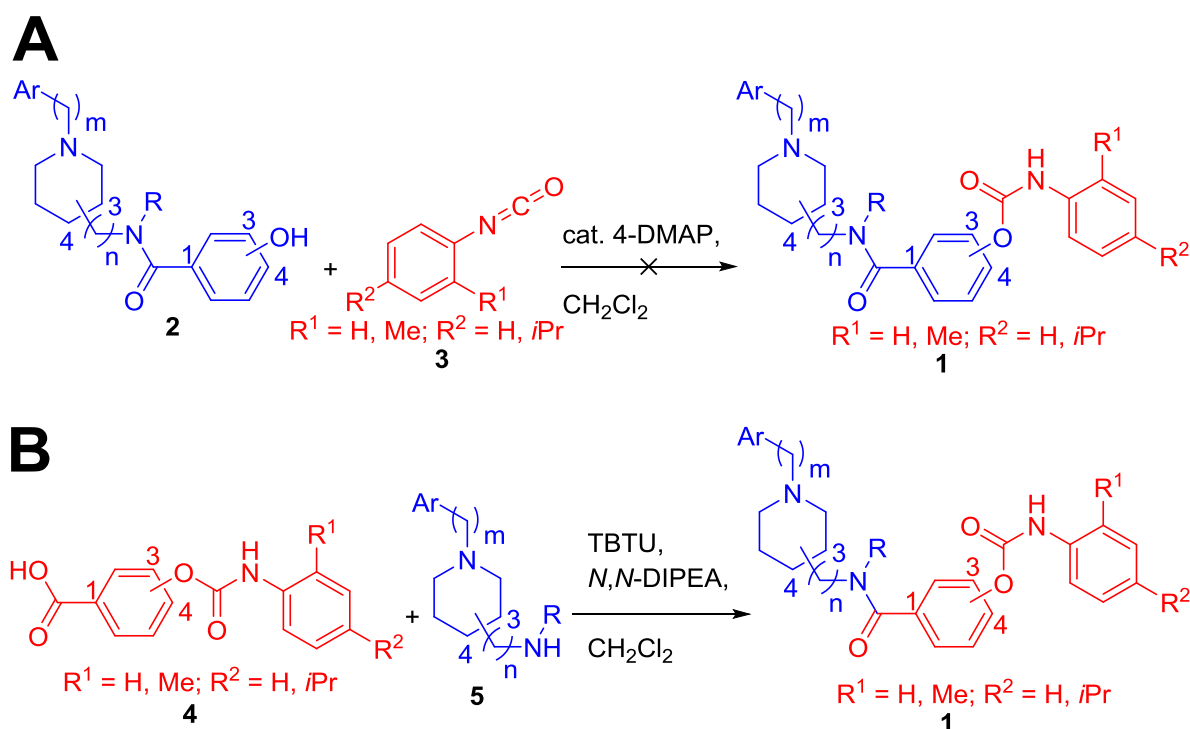
covalent reaction between their carbamate carbonyl group and the catalytic serine in the active site of ChEs (carbamoylation). The inhibited (carbamoylated) ChE is then reactivated by a slow hydrolysis (decarbamylation) of the active enzyme serine (Scheme 1).^{11,12}



Scheme 1. Mechanism of ChE inhibition by phenserine.

As part of our development of new ChE inhibitors as potential anti-Alzheimer disease drugs, we designed compounds with the general formula **1** that contain the phenyl phenylcarbamate moiety (Scheme 1A). We planned to synthesize these compounds by utilizing one of several methods for the synthesis of carbamates,¹³ i.e. reacting phenols with the general formula **2** with various phenyl isocyanates (**3**) in the presence of a catalytic

amount of 4-dimethylamino pyridine (4-DMAP) in CH₂Cl₂ (Scheme 2A).^{14,15} However, this reaction did not produce the desired carbamates. Therefore, we had to plan an alternative synthetic route. We decided to use 3- and 4-((phenylcarbamoyl)oxy)benzoic acids (**4**) and react them with various amines (**5**) in the presence of coupling reagent TBTU and *N,N*-diisopropylethylamine (DIPEA) in CH₂Cl₂¹⁶ to produce the designed amides (Scheme 2B).



Scheme 2. Synthesis of new ChE inhibitors as potential anti-Alzheimer disease drugs with the general formula **1**.

The problem was that 3- and 4-((phenylcarbamoyl)oxy)benzoic acids (**4**; Scheme 1B) are not commercially available and procedures for their preparation have also not been reported yet. Herein we describe how we solved this problem by developing a simple procedure to produce these building blocks in high overall yields.

2. Experimental

2.1. General chemistry methods

^1H -NMR and ^{13}C -NMR were recorded at 400.130 MHz and 100.613 MHz, respectively, on an NMR spectrophotometer (Bruker Avance III). The chemical shifts (δ) are reported in parts per million (ppm) and are referenced to the deuterated solvent used. The coupling constants (J) are reported in Hz, and the splitting patterns are indicated as: s, singlet; bs, broad singlet; d, doublet; dd, doublet of doublets; td, triplet of doublets; h, hextet; m, multiplet; t, triplet; bt, broad triplet; dt, doublet of triplets; tt, triplet of triplets; q, quartet; qd, quartet of doublets. Infrared (IR) spectra were recorded on a FT-IR spectrometer (System Spectrum BX; Perkin-Elmer). ATR IR spectra were recorded on a FT-IR spectrometer (Thermo Nicolet Nexus 470 ESP). Micro-analyses were performed on a Perkin-Elmer C, H, N Analyzer 240 C. The analyses are indicated by the symbols of the elements and they were within $\pm 0.4\%$ of the theoretical values. Mass spectra were recorded on a LC-MS/MS system (Q Executive Plus; Thermo Scientific, MA, USA). Melting points were determined on a Leica hot-stage microscope and are uncorrected. Evaporation of the solvents was performed under reduced pressure. Reagents and solvents were purchased from Acros Organics, Alfa Aesar, Euriso-Top, Fluka, Merck, Sigma-Aldrich, and TCI Europe, and were used without further purification, unless otherwise stated. Flash column chromatography was performed on silica gel 60 for column chromatography (particle size, 230–400 mesh). Analytical thin-layer chromatography was performed on silica gel aluminum sheets (0.20 mm; 60 F254; Merck), with visualization using ultraviolet light and/or visualization reagents. Analytical reversed-phase UPLC method A was performed on an LC system (Dionex Ultimate 3000 Binary Rapid Separation; Thermo Scientific) equipped with an autosampler, a binary pump system, a photodiode array detector, a thermostated column compartment, and the Chromeleon Chromatography Data System. The detector on UPLC system was set to 210 nm and 254 nm.

The column used for method A and was a C18 analytical column (50×2.1 mm, $1.8 \mu\text{m}$; Acquity UPLC HSS C18SB). The column was thermostated at 40°C .

Method A: The sample solution ($1 \mu\text{L}$; 0.2 mg/mL in MeCN) was injected and eluted at a flow rate of 0.4 mL/min , using a linear gradient of mobile phase A (MeCN) and mobile phase B (0.1% [v/v] aqueous TFA). The gradient for method A (for mobile phase A) was: 0–2 min, 20%; 2–5 min, 20–90%; 5–8 min, 90%.

2.2. General synthetic procedures

2.2.1. General procedure for synthesis of benzyl esters (general procedure 1)

To a 100-mL round-bottom flask equipped with a stirring bar, hydroxybenzoic acid (5.000 g , 36.177 mmol , 1.0 mol. equiv.) and DMF (50 mL) were added. The resulting solution was stirred and Na_2CO_3 (3.837 g , 36.177 mmol , 1.0 mol. equiv.) was added. Benzyl bromide (4.297 mL , 36.177 mmol , 1.0 mol. equiv.) was added dropwise to the suspension and the reaction mixture was stirred for 24 hours at room temperature, then poured into a 500-mL separating funnel. Water (100 mL) was added and the mixture was extracted with Et_2O ($3 \times 150 \text{ mL}$). The combined organic phases were transferred into a 1-L separating funnel, washed with water ($3 \times 450 \text{ mL}$) followed by sat. brine solution (450 mL), dried over anhydrous Na_2SO_4 , and evaporated to produce the benzyl hydroxybenzoate as a colourless oil which solidified into a white solid after cooling. This product was used in the next step without further purification.

2.2.2. General procedure for synthesis of carbamates (general procedure 2)

To a round-bottom flask equipped with a stirring bar, benzyl hydroxybenzoate (1.0 mol. equiv.) and CH_2Cl_2 ($c = 0.3 \text{ M}$) were added. The resulting solution was stirred and 4-DMAP

(0.01 mol. equiv.) was added. Phenyl isocyanate, 2-methylphenyl isocyanate or 4-isopropylphenyl isocyanate (1.0 mol. equiv.) was added dropwise and the reaction mixture was stirred for 24 hours at room temperature, then evaporated to produce the carbamates. This product was used in the next step without further purification.

2.2.3. General procedure for debenzylation of benzyl esters (general procedure 3)

To a round-bottom flask equipped with a stirring bar, benzyl ester (1.0 mol. equiv.) and inhibitor free THF ($c = 0.02 \text{ g/mL}$) were added. The resulting solution was stirred and agitated with a stream of argon for 30 min. 10% Pd/C (5% mass of benzyl ester) was added and the resulting suspension was agitated with a stream of hydrogen for 30 min. The reaction mixture was stirred under an atmosphere of hydrogen for 24 hours then agitated with a stream of argon for 30 min, filtered with suction through a pad of Celite and evaporated to produce the carboxylic acid.

2.3. Synthesis and characterization of compounds

2.3.1. Synthesis of benzyl 3-hydroxybenzoate (6)

Synthesized from 3-hydroxybenzoic acid (**7**) (5.000 g, 36.177 mmol, 1.0 mol. equiv.), Na_2CO_3 (3.837 g, 36.177 mmol, 1.0 mol. equiv.) and benzyl bromide (4.297 mL, 36.177 mmol, 0.01 mol. equiv) in DMF (50 mL) via general procedure 1 to produce 7.750 g of **6** as a white solid (94% yield). $R_f = 0.52$ ($\text{CH}_2\text{Cl}_2/\text{MeOH}$, 20:1, v/v). $^1\text{H-NMR}$ (400.130 MHz, CDCl_3): $\delta = 5.17$ (s, 1 H), 5.36 (s, 2 H), 7.05 (dd, $J_1 = 8.0 \text{ Hz}$, $J_2 = 2.4 \text{ Hz}$, 1 H), 7.30–7.45 (m, 6 H), 7.56 (s, 1 H), 7.66 (d, $J = 7.8 \text{ Hz}$, 1 H).

2.3.2. Synthesis of benzyl 4-hydroxybenzoate (8)

Synthesized from 4-hydroxybenzoic acid (**9**) (5.000 g, 36.177 mmol, 1.0 mol. equiv.), Na₂CO₃ (3.837 g, 36.177 mmol, 1.0 mol. equiv.) and benzyl bromide (4.297 mL, 36.177 mmol, 1.0 mol. equiv) in DMF (50 mL) via general procedure 1 to produce 7.073 g of **8** as a white solid (86% yield). *R*_f = 0.46 (CH₂Cl₂/MeOH, 20:1, v/v). ¹H-NMR (400.130 MHz, CDCl₃): δ = 5.34 (s, 2 H), 5.58 (s, 1 H), 6.86 (d, *J* = 8.7 Hz, 2 H), 7.32–7.45 (m, 5 H), 8.00 (d, *J* = 8.7 Hz, 2 H).

2.3.3. Synthesis of benzyl 3-((phenylcarbamoyl)oxy)benzoate (**10**)

Synthesized from **6** (3.249 g, 14.235 mmol, 1.0 mol. equiv.), phenyl isocyanate (1.547 mL, 14.235 mmol, 1.0 mol. equiv.) and 4-DMAP (0.017 g, 0.142 mmol, 0.01 mol. equiv.) in CH₂Cl₂ (47 mL) via general procedure 2 to produce 4.750 g of **10** as a white solid (96% yield). *R*_f = 0.44 (CH₂Cl₂). mp 121–123 °C. IR (ATR): 3319, 1707, 1544, 1440, 1278, 1202, 1107, 732, 692 cm⁻¹. ¹H-NMR (400.130 MHz, CDCl₃): δ = 5.37 (s, 2 H), 6.96 (bs, 1 H), 7.13 (t, *J* = 7.3 Hz, 1 H), 7.33–7.49 (m, 11 H), 7.89 (s, 1 H), 7.97 (d, *J* = 7.6 Hz, 1 H). ¹³C-NMR (100 MHz, DMSO-d₆): δ = 66.41, 118.51, 122.52, 123.06, 126.23, 127.03, 127.97, 128.10, 128.46, 128.81, 129.98, 130.93, 135.90, 138.39, 150.60, 151.35, 164.82. HRMS (ESI⁺): *m/z* calcd for C₂₁H₁₈NO₄: 348.12303; found: 348.12410. Anal. Calcd for C₂₁H₁₇NO₄: C, 72.61; H, 4.93; N, 4.03 Found: C, 72.65; H, 4.96; N, 4.00.

2.3.4. Synthesis of benzyl 3-((*o*-tolylcarbamoyl)oxy)benzoate (**11**)

Synthesized from **6** (3.463 g, 15.172 mmol, 1.0 mol. equiv.), 2-methylphenyl isocyanate (1.881 mL, 15.172 mmol, 1.0 mol. equiv.) and 4-DMAP (0.019 g, 0.152 mmol, 0.01 mol. equiv.) in CH₂Cl₂ (50 mL) via general procedure 2 to produce 5.373 g of **11** as a white solid (98% yield). *R*_f = 0.32 (CH₂Cl₂). mp 78–80 °C. IR (ATR): 3273, 1712, 1531, 1289, 1270, 1232, 1189, 1069, 1022, 747 cm⁻¹. ¹H-NMR (400.130 MHz, CDCl₃): δ = 2.34 (s, 3 H), 5.37

(s, 2 H), 6.75 (bs, 1 H), 7.08 (t, $J = 7.3$ Hz, 1 H), 7.22 (t, $J = 7.6$ Hz, 2 H), 7.33–7.49 (m, 7 H), 7.83 (bs, 1 H), 7.89 (s, 1 H), 7.96 (d, $J = 7.5$ Hz, 1 H). ^{13}C -NMR (100 MHz, DMSO- d_6): $\delta = 17.72, 66.41, 115.66, 119.88, 120.41, 122.42, 126.08, 126.12, 126.93, 127.91, 127.97, 128.02, 128.09, 128.45, 129.92, 130.37, 130.90, 135.65, 135.91, 150.92, 152.33, 164.85$. HRMS (ESI $^{+}$): m/z calcd for $\text{C}_{22}\text{H}_{20}\text{NO}_4$: 362.13868; found: 362.13802. Anal. Calcd for $\text{C}_{22}\text{H}_{19}\text{NO}_4$: C, 73.12; H, 5.30; N, 3.88. Found: C, 73.11; H, 5.26; N, 3.92.

2.3.5. Synthesis of benzyl 3-(((4-isopropylphenyl)carbamoyl)oxy)benzoate (12)

Synthesized from **6** (3.242 g, 14.204 mmol, 1.0 mol. equiv.), 4-isopropylphenyl isocyanate (2.267 mL, 14.204 mmol, 1.0 mol. equiv.) and 4-DMAP (0.017 g, 0.142 mmol, 0.01 mol. equiv.) in CH_2Cl_2 (47 mL) via general procedure 2 to produce 5.278 g of **12** as a white solid (95% yield). $R_f = 0.45$ (CH_2Cl_2). mp 99–101 °C. IR (ATR): 3322, 2963, 1710, 1529, 1445, 1275, 1231, 1100, 741 cm^{-1} . ^1H -NMR (400.130 MHz, CDCl_3): $\delta = 1.24$ (d, $J = 6.8$ Hz, 6 H), 2.84–2.94 (m, 1 H), 5.37 (s, 2 H), 6.92 (bs, 1 H), 7.20 (d, $J = 8.3$ Hz, 2 H), 7.33–7.49 (m, 9 H), 7.88 (s, 1 H), 7.96 (d, $J = 7.5$ Hz, 1 H). ^{13}C -NMR (100 MHz, DMSO- d_6): $\delta = 23.82, 32.74, 66.39, 118.63, 122.46, 126.13, 126.50, 126.98, 127.95, 128.07, 128.43, 129.94, 130.88, 135.88, 136.06, 143.13, 150.65, 151.35, 164.80$. HRMS (ESI $^{+}$): m/z calcd for $\text{C}_{24}\text{H}_{24}\text{NO}_4$: 390.16998; found: 390.16931. Anal. Calcd for $\text{C}_{24}\text{H}_{23}\text{NO}_4$: C, 74.02; H, 5.95; N, 3.60. Found: C, 74.05; H, 5.92; N, 3.58.

2.3.6. Synthesis of benzyl 4-(phenylcarbamoyl)oxy)benzoate (13)

Synthesized from **8** (3.010 g, 13.187 mmol, 1.0 mol. equiv.), phenyl isocyanate (1.433 mL, 13.187 mmol, 1.0 mol. equiv.) and 4-DMAP (0.016 g, 0.132 mmol, 0.01 mol. equiv.) in CH_2Cl_2 (44 mL) via general procedure 2 to produce 4.415 g of **13** as a white solid (96% yield). $R_f = 0.33$ (CH_2Cl_2). mp 103–105 °C. IR (ATR): 3331, 1706, 1543, 1264, 1216, 1102,

1007, 752, 690 cm^{-1} . ^1H -NMR (400.130 MHz, CDCl_3): δ = 5.34 (s, 2 H), 6.95 (bs, 1 H), 7.10 (t, J = 7.3 Hz, 1 H), 7.24 (d, J = 3.4 Hz, 2 H), 7.30–7.43 (m, 9 H), 8.10 (d, J = 8.6 Hz, 2 H). ^{13}C -NMR (100 MHz, DMSO- d_6): δ = 66.16, 115.35, 118.15, 118.55, 122.08, 123.13, 126.58, 127.86, 128.02, 128.43, 128.82, 130.82, 131.50, 136.06, 138.32, 150.93, 154.43, 164.89. HRMS (ESI $^{+}$): m/z calcd for $\text{C}_{21}\text{H}_{18}\text{NO}_4$: 348.12303; found: 348.12249. Anal. Calcd for $\text{C}_{21}\text{H}_{17}\text{NO}_4$: C, 72.61; H, 4.93; N, 4.03. Found: C, 72.64; H, 4.96; N, 4.05.

2.3.7. Synthesis of benzyl 4-((*o*-tolylcarbamoyl)oxy)benzoate (**14**)

Synthesized from **8** (3.453 g, 15.128 mmol, 1.0 mol. equiv.), 2-methylphenyl isocyanate (1.876 mL, 15.128 mmol, 1.0 mol. equiv.) and 4-DMAP (0.018 g, 0.131 mmol, 0.01 mol. equiv.) in CH_2Cl_2 (50 mL) via general procedure 2 to produce 4.975 g of **14** as a white solid (91% yield). R_f = 0.27 (CH_2Cl_2). mp 87–89 $^{\circ}\text{C}$. IR (ATR): 3264, 1705, 1531, 1454, 1272, 1207, 1232, 1016, 753, 696 cm^{-1} . ^1H -NMR (400.130 MHz, CDCl_3): δ = 2.34 (s, 3 H), 5.37 (s, 2 H), 6.76 (bs, 1 H), 7.09 (t, J = 7.2 Hz, 1 H), 7.22 (t, J = 8.1 Hz, 2 H), 7.29 (d, J = 8.6 Hz, 2 H), 7.33–7.45 (m, 5 H), 7.83 (bs, 1 H), 8.12 (d, J = 8.7 Hz, 2 H). ^{13}C -NMR (100 MHz, DMSO- d_6): δ = 17.70, 66.14, 115.34, 121.98, 124.91, 126.14, 126.40, 126.98, 127.77, 127.87, 128.03, 128.39, 128.44, 130.39, 130.79, 131.48, 135.54, 136.06, 136.42, 151.89, 154.74, 164.90. HRMS (ESI $^{+}$): m/z calcd for $\text{C}_{22}\text{H}_{20}\text{NO}_4$: 362.13868; found: 362.13803. Anal. Calcd for $\text{C}_{22}\text{H}_{19}\text{NO}_4$: C, 73.12; H, 5.30; N, 3.88. Found: C, 73.16; H, 5.33; N, 3.91.

2.3.8. Synthesis of benzyl 4-(((4-isopropylphenyl)carbamoyl)oxy)benzoate (**15**)

Synthesized from **8** (2.988 g, 13.091 mmol, 1.0 mol. equiv.), 4-isopropylphenyl isocyanate (2.089 mL, 13.091 mmol, 1.0 mol. equiv.) and 4-DMAP (0.016 g, 0.131 mmol, 0.01 mol. equiv.) in CH_2Cl_2 (44 mL) via general procedure 2 to produce 4.960 g of **15** as a white solid (97% yield). R_f = 0.37 (CH_2Cl_2). mp 112–114 $^{\circ}\text{C}$. IR (ATR): 3329, 2962, 1717,

1537, 1415, 1202, 1113, 1006, 831, 689 cm^{-1} . ^1H -NMR (400.130 MHz, CDCl_3): δ = 1.24 (d, J = 7.0 Hz, 6 H), 2.84–2.95 (m, 1 H), 5.37 (s, 2 H), 6.91 (bs, 1 H), 7.21 (d, J = 8.3 Hz, 2 H), 7.27 (d, J = 8.3 Hz, 2 H), 7.33–7.45 (m, 7 H), 8.12 (d, J = 8.6 Hz, 2 H). ^{13}C -NMR (100 MHz, DMSO- d_6): δ = 23.81, 32.75, 66.13, 115.32, 118.24, 118.67, 122.02, 126.52, 127.84, 128.01, 128.42, 131.47, 136.05, 143.24, 150.92, 154.50, 164.88. HRMS (ESI+): m/z calcd for $\text{C}_{24}\text{H}_{24}\text{NO}_4$: 390.16998; found: 390.17214. Anal. Calcd for $\text{C}_{24}\text{H}_{23}\text{NO}_4$: C, 74.02; H, 5.95; N, 3.60. Found: C, 73.99; H, 5.99; N, 3.57.

2.3.9. Synthesis of 3-((phenylcarbamoyl)oxy)benzoic acid (**16**)

Synthesized from **10** (5.163 g, 14.863 mmol, 1.0 mol. equiv.) and 10% Pd/C (0.258 g, 5% mass of **10**) in inhibitor free THF (258 mL) via general procedure 3 to produce 3.720 g of **16** as a white solid (96% yield). R_f = 0.00 (CH_2Cl_2). mp 151–153 $^\circ\text{C}$. IR (ATR): 3337, 2564, 1685, 1523, 1439, 1302, 1208, 1016, 754 cm^{-1} . ^1H -NMR (400.130 MHz, acetone- d_6): δ = 7.09 (t, J = 7.3 Hz, 1 H), 7.35 (t, J = 7.9 Hz, 2 H), 7.50 (d, J = 8.3 Hz, 1 H), 7.57 (t, J = 7.9 Hz, 1 H), 7.63 (d, J = 8.1 Hz, 2 H), 7.86 (s, 1 H), 7.93 (d, J = 7.6 Hz, 1 H), 9.25 (s, 1 H), 11.43 (bs, 1 H). ^{13}C -NMR (100 MHz, acetone- d_6): δ = 120.50, 124.77, 125.13, 128.32, 128.34, 130.74, 131.35, 133.91, 140.54, 152.98, 153.37, 167.96. HRMS (ESI+): m/z calcd for $\text{C}_{14}\text{H}_{12}\text{NO}_4$: 258.07608; found: 258.07740. UPLC purity, 99% at 254 nm (method A, t_R = 4.130 min).

2.3.10. Synthesis of 3-((*o*-tolylcarbamoyl)oxy)benzoic acid (**17**)

Synthesized from **11** (5.292 g, 14.643 mmol, 1.0 mol. equiv.) and 10% Pd/C (0.265 g, 5% mass of **11**) in inhibitor free THF (265 mL) via general procedure 3 to produce 3.902 g of **17** as a white solid (98% yield). R_f = 0.00 (CH_2Cl_2). mp 176–178 $^\circ\text{C}$. IR (ATR): 3298, 2565, 1686, 1530, 1449, 1306, 1221, 1023, 942, 750 cm^{-1} . ^1H -NMR (400.130 MHz, acetone- d_6): δ

= 2.39 (s, 3 H), 7.11 (t, J = 7.5 Hz, 1 H), 7.19–7.26 (m, 2 H), 7.49 (d, J = 8.3 Hz, 1 H), 7.56 (t, J = 7.9 Hz, 1 H), 7.63 (d, J = 7.9 Hz, 1 H), 7.85 (s, 1 H), 7.92 (d, J = 7.6 Hz, 1 H), 8.50 (bs, 1 H), 11.32 (bs, 1 H). ^{13}C -NMR (100 MHz, DMSO- d_6): δ = 17.76, 122.54, 124.87, 125.48, 126.18, 126.34, 129.72, 130.43, 132.07, 132.28, 135.71, 150.85, 152.45, 166.64. HRMS (ESI $^{+}$): m/z calcd for $\text{C}_{15}\text{H}_{14}\text{NO}_4$: 272.09173; found: 272.09150. UPLC purity, 96% at 254 nm (method A, t_R = 4.193 min).

2.3.11. Synthesis of 3-(((4-isopropylphenyl)carbamoyl)oxy)benzoic acid (**18**)

Synthesized from **12** (5.193 g, 13.334 mmol, 1.0 mol. equiv.) and 10% Pd/C (0.260 g, 5% mass of **12**) in inhibitor free THF (260 mL) via general procedure 3 to produce 3.795 g of **18** as a white solid (95% yield). R_f = 0.00 (CH_2Cl_2). mp 174–176 °C. IR (ATR): 3320, 2961, 2541, 1715, 1682, 1538, 1450, 1274, 1225, 1017, 840 cm^{-1} . ^1H -NMR (400.130 MHz, acetone- d_6): δ = 1.23 (d, J = 7.0 Hz, 6 H), 2.84–2.94 (m, 1 H), 7.23 (d, J = 8.4 Hz, 2 H), 7.48 (m, 4 H), 7.85 (s, 1 H), 7.92 (d, J = 7.7 Hz, 1 H), 9.18 (s, 1 H), 11.41 (bs, 1 H). ^{13}C -NMR (100 MHz, acetone- d_6): δ = 25.35, 35.19, 120.65, 124.76, 128.28, 128.31, 128.53, 131.32, 133.82, 138.18, 145.68, 153.03, 153.39, 167.98. HRMS (ESI $^{+}$): m/z calcd for $\text{C}_{17}\text{H}_{18}\text{NO}_4$: 300.12303; found: 300.12463. UPLC purity, 99% at 254 nm (method A, t_R = 4.723 min).

2.3.12. Synthesis of 4-((phenylcarbamoyl)oxy)benzoic acid (**19**)

Synthesized from **13** (4.334 g, 12.477 mmol, 1.0 mol. equiv.) and 10% Pd/C (0.217 g, 5% mass of **13**) in inhibitor free THF (217 mL) via general procedure 3 to produce 3.194 g of **19** as a white solid (99% yield). R_f = 0.00 (CH_2Cl_2). mp 195–197 °C. IR (ATR): 3305, 2557, 1682, 1527, 1502, 1427, 1292, 1198, 1012, 752 cm^{-1} . ^1H -NMR (400.130 MHz, acetone- d_6): δ = 7.10 (t, J = 7.3 Hz, 1 H), 7.34–7.38 (m, 4 H), 7.63 (d, J = 7.9 Hz, 2 H), 8.10 (d, J = 8.1 Hz, 2 H), 9.28 (s, 1 H), 11.10 (bs, 1 H). ^{13}C -NMR (100 MHz, DMSO- d_6): δ = 115.07, 118.53,

121.89, 123.14, 127.90, 128.87, 130.82, 131.49, 138.33, 151.04, 153.98, 166.67. HRMS (ESI+): m/z calcd for $C_{14}H_{12}NO_4$: 258.07608; found: 258.07647. UPLC purity, 98% at 254 nm (method A, t_R = 4.153 min).

2.3.13. Synthesis of 4-((*o*-tolylcarbamoyl)oxy)benzoic acid (**20**)

Synthesized from **14** (4.640 g, 12.839 mmol, 1.0 mol. equiv.) and 10% Pd/C (0.232 g, 5% mass of **14**) in inhibitor free THF (232 mL) via general procedure 3 to produce 3.384 g of **20** as a white solid (97% yield). R_f = 0.00 (CH_2Cl_2). mp 183–185 °C. IR (ATR): 3280, 2555, 1685, 1529, 1426, 1291, 1234, 1208, 1161, 748 cm^{-1} . 1H -NMR (400.130 MHz, acetone- d_6): δ = 2.39 (s, 3 H), 7.11 (t, J = 7.5 Hz, 1 H), 7.20–7.27 (m, 2 H), 7.36 (d, J = 8.6 Hz, 2 H), 7.62 (d, J = 7.7 Hz, 1 H), 8.09 (d, J = 8.6 Hz, 2 H), 8.53 (bs, 1 H), 11.14 (bs, 1 H). ^{13}C -NMR (100 MHz, DMSO- d_6): δ = 17.77, 115.14, 121.81, 124.99, 125.56, 126.61, 127.80, 130.46, 130.86, 131.55, 135.63, 152.08, 154.37, 166.75. HRMS (ESI+): m/z calcd for $C_{15}H_{14}NO_4$: 272.09173; found: 272.09436. UPLC purity, 97% at 254 nm (method A, t_R = 4.213 min).

2.3.14. Synthesis of 4-(((4-isopropylphenyl)carbamoyl)oxy)benzoic acid (**21**)

Synthesized from **15** (4.864 g, 12.489 mmol, 1.0 mol. equiv.) and 10% Pd/C (0.243 g, 5% mass of **15**) in inhibitor free THF (243 mL) via general procedure 3 to produce 3.720 g of **21** as a white solid (99% yield). R_f = 0.00 (CH_2Cl_2). mp 196–198 °C. IR (ATR): 3362, 2964, 2547, 1720, 1676, 1501, 1198, 1011, 828, 758 cm^{-1} . 1H -NMR (400.130 MHz, acetone- d_6): δ = 1.23 (d, J = 6.8 Hz, 6 H), 2.84–2.94 (m, 1 H), 7.24 (d, J = 8.3 Hz, 2 H), 7.36 (d, J = 8.4 Hz, 2 H), 7.53 (d, J = 8.2 Hz, 2 H), 8.09 (d, J = 8.6 Hz, 2 H), 9.19 (s, 1 H), 11.19 (bs, 1 H). ^{13}C -NMR (100 MHz, DMSO- d_6): δ = 23.90, 32.83, 118.75, 121.85, 126.60, 127.87, 130.87, 136.08, 143.31, 151.12, 154.14, 166.75. HRMS (ESI+): m/z calcd for $C_{17}H_{18}NO_4$: 300.12303; found: 300.12476. UPLC purity, 99% at 254 nm (method A, t_R = 4.743 min).

295
296 **2.3.15. Synthesis of 4-((1-(2,3-dihydro-1H-inden-2-yl)piperidin-3-yl)carbamoyl)phenyl**
297 **phenylcarbamate (22)**

298 To a 50-mL round-bottom flask equipped with a stirring bar, compound **16** (0.100 g,
299 0.389 mmol, 1.0 equiv) was added followed by CH₂Cl₂ (10 mL). The resulting suspension
300 was stirred and cooled to 0 °C. *N,N*-diisopropylethylamine (0.135 mL, 0.778 mmol, 2.0
301 equiv) was added dropwise and the suspension transformed into a solution. TBTU was added
302 and 30 min later solution A (see below) was added dropwise. The reaction mixture was
303 allowed to warm to room temperature and then stirred for 24 hours. During this time a white
304 precipitate formed. The suspension was filtered with suction to produce 0.133 g of compound
305 **5** as a white solid (75% yield).

306 Preparation of solution A: To 25-mL round-bottom flask equipped with a stirring bar,
307 compound **23** (0.112 g, 0.389 mmol, 1.0 equiv) was added followed by CH₂Cl₂ (11 mL). The
308 resulting suspension was stirred and cooled to 0 °C. *N,N*-diisopropylethylamine (0.135 mL,
309 0.778 mmol, 2.0 equiv) was added dropwise and the suspension transformed into a solution.

310 Characterization of compound 22: *R*_f = 0.46 (CH₂Cl₂/MeOH, 10:1, v/v). mp 137–139 °C.
311 IR (ATR): 3293, 2953, 1716, 1629, 1538, 1226, 1211, 1021, 694 cm⁻¹. ¹H-NMR (400.130
312 MHz, DMSO-d₆): δ = 1.33–1.43 (m, 1 H), 1.48–1.60 (m, 1 H), 1.73 (d, *J* = 13.3 Hz, 1 H),
313 1.83 (d, *J* = 10.5 Hz, 1 H), 1.96–2.01 (m, 2 H), 2.75–2.86 (m, 3 H), 2.98–3.05 (m, 3 H), 3.20
314 (t, *J* = 7.7 Hz, 1 H), 3.97 (bs, 1 H), 7.04–7.12 (m, 3 H), 7.17–7.18 (m, 2 H), 7.33 (t, *J* = 7.7
315 Hz, 2 H), 7.39 (d, *J* = 8.4 Hz, 1 H), 7.50–7.53 (m, 3 H), 7.70 (s, 1 H), 7.76 (d, *J* = 7.7 Hz, 1
316 H), 8.27 (d, *J* = 7.7 Hz, 1 H), 10.29 (s, 1 H). ¹³C-NMR (100 MHz, DMSO-d₆): δ = 23.75,
317 29.80, 36.19, 36.28, 46.47, 50.70, 55.99, 66.07, 118.42, 120.79, 123.00, 124.17, 124.44,
318 124.72, 126.19, 128.81, 129.27, 135.81, 138.42, 141.27, 150.28, 151.54, 164.55. HRMS

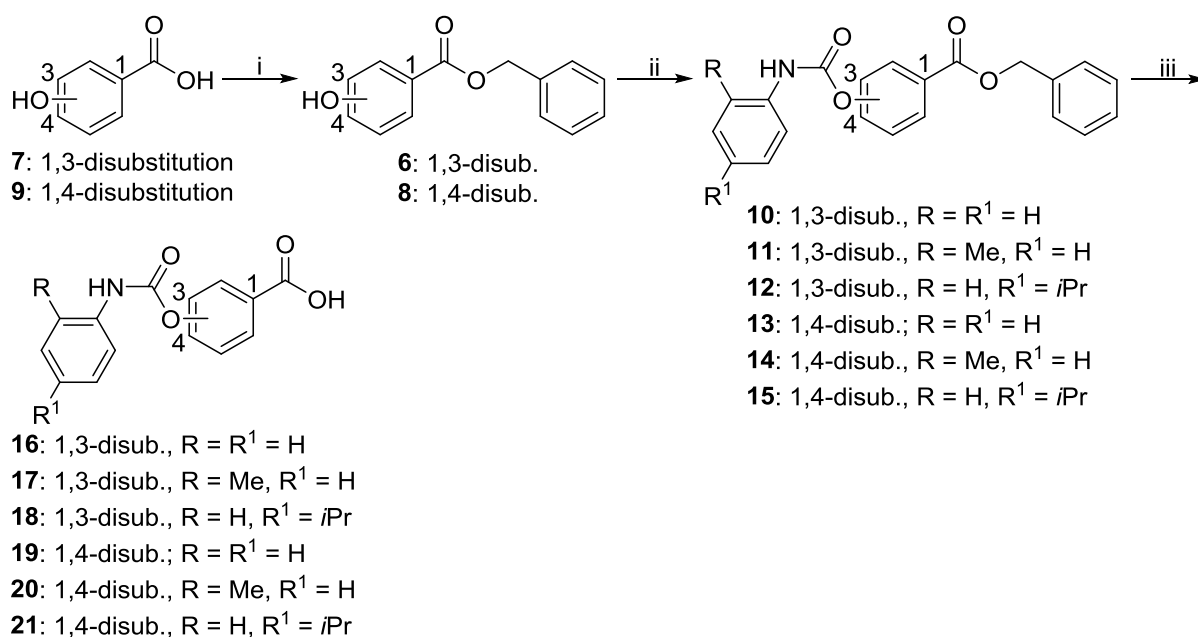
(ESI+): m/z calcd for C₂₈H₃₀N₃O₃: 456.22817; found: 456.22717. UPLC purity, 96% at 254 nm (method A, t_R = 4.420 min).

3. Results and Discussion

For the synthesis of 3-((phenylcarbamoyl)oxy)benzoic acid (**16**), commercially available 3-hydroxybenzoic acid (**7**) was treated with benzyl bromide in the presence of Na₂CO₃ in DMF,¹⁷ to provide benzyl 3-hydroxybenzoate (**6**) in 86% yield. No further purification of compound **6** was required and the diethyl ether used for the extraction of compound **6** was reused for the extraction in the synthesis of benzyl 4-hydroxybenzoate (**8**) (Scheme 3).

In the second step, compound **6** was converted into carbamate **10** with one equivalent of phenyl isocyanate in the presence of a catalytic amount (0.01 equivalent) of 4-DMAP in CH₂Cl₂^{14,15} in 96% yield. Again, no further purification of carbamate **10** was required. Using one equivalent of phenyl isocyanate, rather than 1.10¹⁵ or 1.20 equivalent,¹⁴ was found to be an advantage as no over-reaction occurred. As reported previously, excess phenyl isocyanate can undergo an ArS_E substitution in the phenyl moiety of the carbamate to produce an amide, which can be difficult to separate from the desired carbamate.¹⁵ Additionally, 1.0 mol% rather than 5 mol%^{14,15} of 4-DMAP was enough to produce the desired carbamate in excellent yield (Scheme 3).

In the third and final step, the benzyl ester **10** was debenzylated using classic catalytic hydrogenation with gaseous hydrogen and a catalytic amount of 10% Pd/C¹⁸ (5% mass of benzyl ester **10**) in inhibitor free THF to produce carboxylic acid **16** in 99% yield (Scheme 3). The hydrogenation was a very clean reaction: no further purification of acid **16** was required and the inhibitor free THF was reused for the debenzylation of benzyl esters **11–15**.

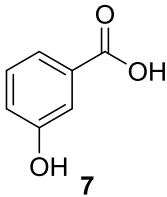
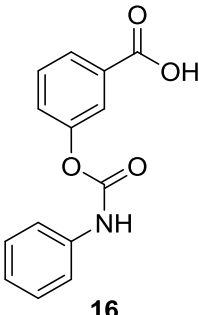
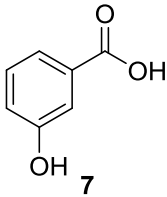
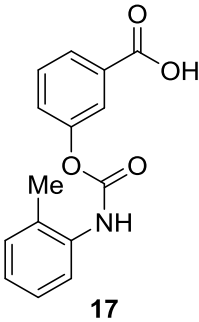
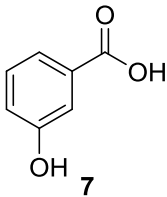
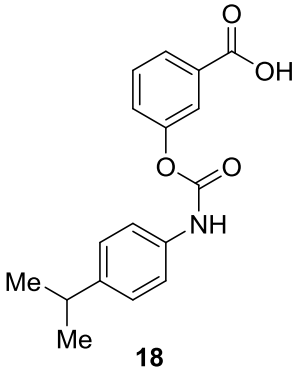
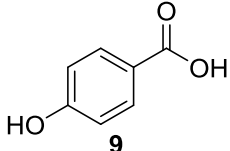
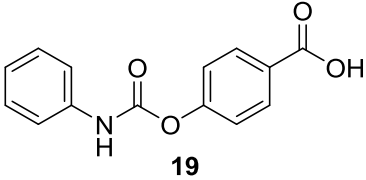
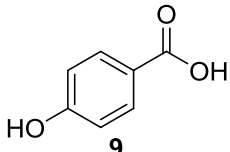
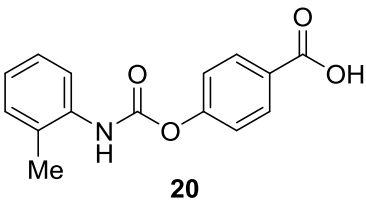
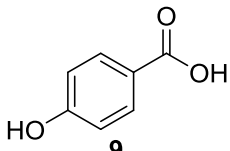
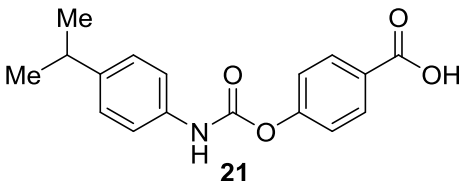


Scheme 3. Reagents and conditions: (i) PhCH₂Br, Na₂CO₃, DMF, rt, 24 h, 94% (for 8) and 86% (for 9); (ii) phenyl isocyanate, 4-DMAP, CH₂Cl₂, rt, 24 h, 91–98%; (iii) H₂(g), 10% Pd/C, THF, rt, 24 h, 95–99%.

The overall yield for the preparation of compound **16** from 3-hydroxybenzoic acid (**7**) using this procedure was 87% (Table 1). The same procedure was then used to prepare compounds **17–21** from the corresponding hydroxybenzoic acids **7** or **9** (Scheme 3). Overall yields ranged from 76–90% and are reported in Table 1.

Table 1. The synthesized 3- and 4-((phenylcarbamoyl)oxy)benzoic acids.

Starting hydroxybenzoic acid	Final ((phenylcarbamoyl)oxy)benzoic acid	Overall yield %
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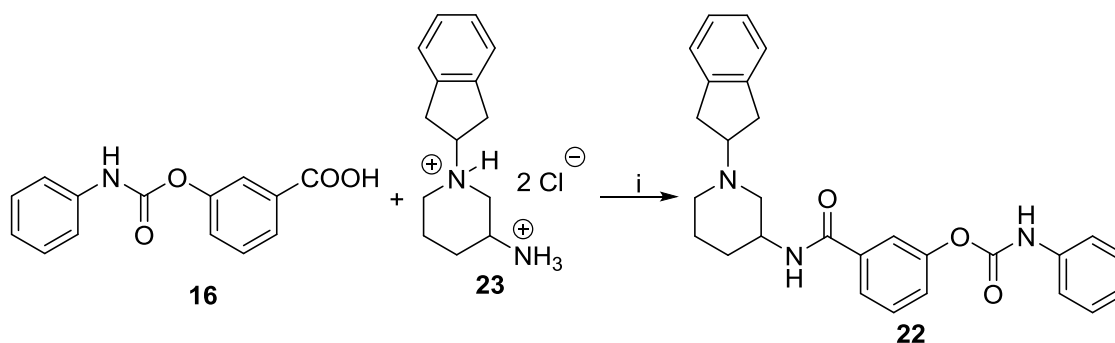
 7	 16	87
 7	 17	90
 7	 18	85
 9	 19	82
 9	 20	76
 9	 21	83

355

356 As a proof of concept that the synthesized 3- and 4-((phenylcarbamoyl)oxy)benzoic acids

357 can be used in the next reaction to prepare amides, carboxylic acid **16** was reacted with amine

23 in the presence of coupling reagent TBTU and *N,N*-diisopropylethylamine (*N,N*-DIPEA) in CH₂Cl₂¹⁶ to produce amide **22** in 75% yield (Scheme 4).



Scheme 4. Reagents and conditions: (i) TBTU, *N,N*-DIPEA, CH₂Cl₂, 0°C to rt, 24 h, 75%.

4. Conclusions

In summary, we have developed method for the synthesis of previously unreported 3- and 4-((phenylcarbamoyl)oxy)benzoic acids from commercially available 3- and 4-hydroxybenzoic acids, respectively. The main advantages of our method are the simplicity, as no purification of intermediates or final acids is required, and effectiveness, as the overall yields are very good to excellent (76–90%). As we have shown, the synthesized carboxylic acids can be converted further, e.g. reacted with amines to produce amides with potential application in drug discovery.

5. Acknowledgements

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