Scientific paper

Synthesis, Antimicrobial Activity and *in silico* Studies on Thymol Esters

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Abstract

Derivatisation of parent structure in terpenoids often results in enhancement of biological activity of newly obtained compounds. Thymol, a naturally occurring phenol biosynthesized through the terpene pathway, is a well known biocide with strong antimicrobial attributes and diverse therapeutic activities. We have aimed our study on a single modification of phenolic functionality in thymol in order to obtain a small focused library of twenty thymyl esters, ten of which were new compounds. All compounds were involved in *in vitro* antimicrobial testing. Another important aspect of current study was implementation of *in silico* calculation of physico-chemical, pharmacokinetic and toxicological properties, which could be helpful by giving an additional guidance in further research.

Keywords: chemical synthesis, thymyl esters, in vitro antimicrobial activity, in silico calculation

1. Introduction

Terpenoids constitute an abundant and potent group of natural products, which play an important role in the enzyme systems of plants and reflect conspicuous biological activity against various pests. Their biological activity is believed to be related to the nature and the position of functional groups or substituents. Chemical modification of natural monoterpenoids has been reported to result in enhancement of biological activities when compared to parent compounds. Thymol (1), a monoterpene biosynthetically directly related to *para*-cymene (1a)⁶ and a naturally occurring phenol, is a well known biocide and most dominant constituent of the oils of thyme (*Thymus vulgaris* L.) and oregano (*Origanum vulgare* L.).

Numerous studies have demonstrated the antimicrobial effect of thymol, ranging from inducing antibiotic

Sheme 1. Chemical structures of thymol (1) and para-cymene (1a).

susceptibility in drug-resistant pathogens through a synergistic effect, to a powerful antioxidant properties.⁷ Thymol has been shown to be an effective fungicide, particularly against fluconazole-resistant strains⁸ and was demonstrated to have strong antimutagenic effect.⁹ In addi-

tion, there is evidence that thymol has antitumor properties. 10

The primary mode of antibacterial action of thymol is not fully known, but is believed to involve outer- and inner membrane disruption, and interaction with membrane proteins and intracellular targets. ^{11–16} The mode of action of thymol against yeast and fungi has been sparsely investigated, though certain studies point to interactions with the cell envelope and intracellular targets. ^{16–19}

Thymyl derivatives are well represented in Asteraceae plants, particularly within tribes Senecioneae, Eupatorieae, Inuleae and Helenieae. In some *Inula*, *Doronicum* and *Pulicaria* species, thymyl derivatives rather than sesquiterpenoids are the major root constituents.^{20–23} A review by Talavera-Aleman and collaborators²⁴ estimated that only 10% of known functionalized thymyl derivatives have been employed in biological testing, showing vast array of diverse activities, such as antimicrobial^{4,5,22,25–27} (several papers reporting inhibitory activity against plants' pathogenic fungi^{22,26}), antioxidant,²⁸ antinociceptive,²⁹ anti-parasitic (antileishmanial),^{30,31} antiprotozoal,³² insecticidal³³ and piscicidal³⁴ activity. The usefulness of thymyl derivatives as transdermal drug delivery enhancers has also been reported.³⁵

There are several papers reporting isolation, synthesys^{4,23,29,33,36} and biological activity^{4,5,22,25,28–30,33} of thymyl esters. Grodnitzky and Coats³³ have tested insecticidal activity of thymyl esters of acetic, dichloracetic, trichloroacetic, chlorodifluoroacetic, pivalic and chloropivalic acid on Musca domestica. Mathela and collaborators⁴ evaluated antibacterial activity of thymyl esters of acetic, propanoic, 2-methylpropanoic, 3-methylbutanoic, but-2-enoic, benzoic and 2-phenylacetic acids and reported the enhancement in the activity of derivatives in comparison to thymol. Kumbhar and Dewang⁵ tested antifungal activity and observed structure-activity relationship, stating that thymyl ethers showed better antifungal potency over esters (thymyl acetate, benzoate, cinnamate, dithymyl malonate, succinate and glutarate) and that the addition of a methylene group or a carbon, an olefinic bond or an aromatic moiety in side chain led to compounds with improved potency over the parent compound (thymol).⁵ Angeles-Lopez et al.²⁹ have undertaken a study to establish the potential acute toxicity and the antinociceptive activity in animal models of thymyl esters of C₂-C₆ straight chain, C₄-C₆ acids positional isomers, diasteroisomers of 2-pentenoic, and of benzoic acid, reporting several esters with antinociceptive effect at a dose of 1 mg/kg.

Keeping in view diverse pharmacological activities of thymol^{7–19} and preliminary results on bioactivity of thymyl derivatives^{4,5,22,25,28–30,33} we have set the aim of our study. By making a single modification of a phenolic functionality in thymol we have obtained a series of ester compounds (**3a–t**), performed their structural characterization, *in vitro* antimicrobial testing and *in silico* calcula-

tion of physico-chemical, pharmacokinetic and toxicological properties. The most important contribution of this study is the synthesis of ten new compounds (**3i**, **3k**–**s**) followed by results obtained in antimicrobial assay and *in silico* calculations, which all together could be an important aspect and an additional guidance in further research.

2. Materials and Methods

2. 1. Chemicals Used

All of the reagents, standards and solvents used were of analytical reagent grade. Unless specified otherwise, all chemicals were purchased from Merck (Darmstadt, Germany).

2. 2. General Synthetic Procedures

Acetyl, benzoyl and palmitoyl chloride were purchased from Sigma–Aldrich. The conversion of the other carboxylic acids to acyl chlorides^{37,38} and the preparation of thymyl esters utilized methods from the literature.^{39,40} Scheme 1 presents the synthesis of thymyl esters.

2. 2. 1. General Procedures for Synthesis of Acid Chlorides

(a) An old procedure developed by Brown³⁷ was used for the preparation of volatile acid chlorides. This procedure involves the action of a relatively slightly volatile benzoyl chloride upon an organic acid. Following this protocol, the synthesis of chlorides up to 10 carbons, as well as 2-chloroacetyl chloride was achieved. In a round-bottom flask equipped by a fractionating column 0.25 mol of the acid and 0.375 mol of benzoyl chloride were placed The mixture was heated until the boiling point was reached, and then the acid chloride was distilled from the reaction mixture. The material so obtained was used directly in the synthesis.

(b) For the preparation of chlorides higher than $10\,\mathrm{C}$ atoms the corresponding acid was refluxed for $2\,\mathrm{h}$ with thionyl chloride in $\mathrm{CCl_4}$. The solvent and the excess of thionyl chloride were removed with the aid of the water pump vacuum.

The products obtained in either way were used directly in the synthesis.

2. 2. 2. General Procedure for the Synthesis of Thymyl Esters 3a-t

Series of thymyl esters were made following the procedure described by Paolini et al. ³⁹ A solution of acid chloride (4.5 mmol) in CH_2Cl_2 (15 mL) was added drop by drop to a cooled mixture (0 °C) of thymol (3.3 mmol) and Et_3N (4.5 mmol) in CH_2Cl_2 (20 mL). The mixture was stirred at room temperature and then refluxed for 3 h. The

organic layer was washed with water ($3 \times 200 \text{ mL}$), dried over anhydrous Na₂SO₄ and concentrated under vacuum. The esters were purified by column chromatography, stationary phase Silica Gel 60 (70–230 mesh), mobile phase (hexane/diethyl ether, gradient 9:1 to 8:2). For yields, see supplementary data.

Thymyl Acetate (3a)

Chromatographic purification gave colorless oil. $C_{12}H_{16}O_2$ (M=192.25); yield 85%; MS (EI): m/z (%) 192 (M⁺) (11.3), 151 (4.8), 150 (42.6), 151 (4.8), 135 (100), 136 (11.0), 115 (7.3), 105 (4.7), 77 (5.1), 43 (4.1); RI (HP5-MS): 1367; 1H NMR (500.13 MHz, CDCl₃): δ (ppm) 7.22 (1H, d, J=8 Hz, Ar-H), 7.05 (1H, d, J=8 Hz, Ar-H), 6.84 (1H, s, Ar-H), 3.00 (1H, spt, J=7 Hz, CH), 2.34 (6H, s, CH₃), 1.22 (6H, d, J=7 Hz, CH₃); 13 C NMR (125.76 MHz, CDCl₃): δ (ppm) 169.77 (C=O), 147.84 (C_{Ar}), 136.95 (C_{Ar}), 136.52 (C_{Ar}), 127.52 (C_{Ar}), 126.78 (C_{Ar}), 123.23 (C_{Ar}), 26.89 (CH), 26.57 (CO-CH₃), 22.99 (2 × CH₃), 21.21 (CH₃).

Thymyl 2-Chloroacetate (3b)

Chromatographic purification gave colorless oil. $C_{12}H_{15}O_2Cl\ (M=226.70);$ yield 85%; MS (EI): $m/z\ (\%)$ 226 (M⁺) (10.1), 150 (30.4), 149 (19.4), 136 (9.4), 133 (11.1), 115 (6.5), 105 (8.1), 91 (14.6), 77 (11.4); RI (HP5-MS): 1572; ¹H NMR (500.13 MHz, CDCl₃): δ (ppm) 7.24 (1H, d, J=8 Hz, Ar-H), 7.08 (1H, d, J=7.6 Hz, Ar-H), 6.87 (1H, s, Ar-H), 4.33 (2H, s, CH₂), 3.00 (1H, spt, J=6.90 Hz, CH), 2.34 (3H, s, CH₃), 1.22 (6H, d, J=6.90 Hz, CH₃). ¹³C NMR (125.76 MHz, CDCl₃): δ (ppm) 166.11 (C=O), 147.45 (C_{Ar}), 136.81 (C_{Ar}), 136.79 (C_{Ar}), 127.64 (C_{Ar}), 126.60 (C_{Ar}), 122.20 (C_{Ar}), 40.74 (CH₂), 27.01 (CH), 22.98 (2 × CH₃), 20.77 (CH₃-Ar).

Thymyl Propanoate (3c)

Chromatographic purification gave colorless oil. $C_{13}H_{18}O_2$ (M=206.28); yield 82%; MS (EI): m/z (%) 206 (M⁺) (9.5), 151 (4.5), 150 (42.1), 135 (100), 136 (9.5), 115 (5.7), 105 (5.4), 91 (11.0), 77 (5.1), 57 (12.6); RI (HP5-MS): 1455; 1H NMR (500.13 MHz, CDCl₃): δ (ppm) 7.21 (1H, d, J=8 Hz, Ar-H), 7.04 (1H, d, J=8 Hz, Ar-H), 6.83 (1H, s, Ar-H), 2.99 (1H, spt, J=14 Hz, CH), 2.63 (2H, q, J=8 Hz, CH₂), 2.33 (3H, s, CH₃), 1.32 (3H, m, CH₃), 1.21 (6H, d, J=7 Hz, CH₃); 13 C NMR (125.76 MHz, CDCl₃): δ (ppm) 173.16 (C=O), 147.93 (C_{Ar}), 136.96 (C_{Ar}), 136.5 (C_{Ar}), 126.69 (C_{Ar}), 123.24 (C_{Ar}), 122.77 (C_{Ar}), 27.73 (CH), 26.91 (CH₂), 22.90 (CH₃), 23.03 (2 × CH₃), 9.31 (CH₃).

Thymyl Butanoate (3d)

Chromatographic purification gave yellowish oil. $C_{14}H_{20}O_2$ (M=220.31); yield 84%; MS (EI): m/z (%) 220 (M⁺) (9.7), 151 (6.1), 150 (55.5), 136 (9.5), 135 (100), 115 (5.4), 105 (5.7), 91 (10.4), 71 (9.3), 43 (7.1); RI (HP5-MS): 1544; ¹H NMR (500.13 MHz, CDCl₃): δ

(ppm) 7.21 (1H, d, J = 8 Hz, Ar-H), 7.04 (1H, d, J = 8 Hz, Ar-H), 6.82 (1H, s, Ar-H), 2.99 (1H, spt, J = 7 Hz, CH), 2.58 (2H, t, J = 7 Hz, CH₂), 2.33 (3H, s, CH₃), 1.84 (2H, sxt, J = 7 Hz, CH₂), 2.1 (6H, d, J = 7 Hz, CH₃), 1.08 (3H, t, J = 7 Hz, CH₃); 13 C NMR (125.76 MHz, CDCl₃): δ (ppm) 172.34 (C=O), 147.88 (C_{Ar}), 136.98 (C_{Ar}), 136.49 (C_{Ar}), 126.66 (C_{Ar}), 125.99 (C_{Ar}), 123.24 (C_{Ar}), 36.24 (CH₂), 26.90 (CH), 23.07 (2 × CH₃), 18.55 (CH₂), 13.78 (CH₃).

Thymyl 2-Methylpropanoate (3e)

Chromatographic purification gave colorless oil, $C_{14}H_{20}O_2$ (M=220.31); yield 75%; MS (EI): m/z (%) 220 (M⁺) (11.4), 150 (59.6), 136 (9.4), 135 (100), 115 (7.3), 105 (8.0), 91 (15.1), 77 (6.8), 71 (18.8), 43 (21.9); RI (HP5-MS): 1495; 1H NMR (500.13 MHz, CDCl₃): δ (ppm) 7.21 (1H, d, J=8 Hz, Ar-H), 7.03 (1H, d, J=7.6 Hz, Ar-H), 6.81 (1H, s, Ar-H), 2.99 (1H, spt, J=6.9 Hz, CH), 2.85 (1H, spt, J=7 Hz, CH), 2.33 (3H, s, CH₃) 1.36 (6H, d, J=7.3 Hz, $Z\times CH_3$), 1.21 (6H, d, J=6.9 Hz, $Z\times CH_3$). ^{13}C NMR (125.76 MHz, CDCl₃): δ (ppm) 175.71 (C=O), 147.99 (C_{Ar}), 136.97 (C_{Ar}), 136.47 (C_{Ar}), 126.62 (C_{Ar}), 126.56 (C_{Ar}), 122.14 (C_{Ar}), 33.95 (CH), 26.78 (CH), 22.99 (2 × CH₃), 19.20 (CH₃), 19.05 (2 × CH₃).

Thymyl Pentanoate (3f)

Chromatographic purification gave colorless oil. $C_{15}H_{22}O_2$ (M=234.33); yield 87%; MS (EI): m/z (%) 234 (M⁺) (8.1), 151 (7.4), 150 (67.6), 135 (100), 136 (9.3), 115 (5.5), 105 (6.1), 91 (10.7), 85 (8.0), 57 (16.4); RI (HP5-MS): 1640; 1H NMR (500.13 MHz, CDCl₃): δ (ppm) 7.21 (1H, d, J=8 Hz, Ar-H), 7.04 (1H, d, J=8 Hz, Ar-H), 6.82 (1H, s, Ar-H), 2.99 (1H, spt, J=14 Hz, CH), 2.6 (2H, t, J=7.6 Hz, CH₂), 2.33 (3H, s, CH₃), 1.79 (2H, quin, J=7.5 Hz, CH₂), 1.49 (2H, sxt, J=15 Hz, CH₂), 1.21 (6H, d, J=6.9 Hz, $2\times CH_3$), 1.01 (3H, t, J=7.5 Hz, CH₃); 13 C NMR (125.76 MHz, CDCl₃): δ (ppm) 172.45 (C=O), 147.90 (C_{Ar}), 136.95 (C_{Ar}), 136.45 (C_{Ar}), 126.96 (C_{Ar}), 126.3 (C_{Ar}), 122.7 (C_{Ar}), 34.06 (CH₂), 27.06 (CH₂), 27.02 (CH), 22.28 (CH₂), 22.96 (2 × CH₃), 20.76 (CH₃-Ar), 18.55 (CH₂), 13.8 (CH₃).

Thymyl 3-Methylbutanoate (3g)

Chromatographic purification gave colorless oil. $C_{15}H_{22}O_2$ (M=234.33); yield 81%; MS (EI): m/z (%) 234 (M⁺) (8.9), 151 (8.3), 150 (71.4), 135 (100), 136 (9.3), 115 (5.4), 105 (6.8), 91 (11.0), 85 (9.3), 52 (21.5); RI (HP5-MS): 1590; 1H NMR (500.13 MHz, CDCl₃): δ (ppm) 7.21 (1H, d, J=8 Hz, Ar-H), 7.04 (1H, d, J=7.6 Hz, Ar-H), 6.81 (1H, s, Ar-H), 3.00 (1H, spt, J=6.9 Hz, CH), 2.48 (2H, d, J=6.9 Hz, CH₂), 2.33 (3H, s, CH₃), 2.28 (1H, m, CH), 1.49 (2H, sxt, J=15 Hz, CH₂), 1.20 (6H, d, J=6.9 Hz, $2 \times CH_3$); ^{13}C NMR (125.76 MHz, CDCl₃): δ (ppm) 171.74 (C=O), 147.87 (C_{Ar}), 136.99 (C_{Ar}), 136.45 (C_{Ar}), 126.99 (C_{Ar}), 126.32 (C_{Ar}), 122.70 (C_{Ar}), 43.34 (CH₂), 26.98

(CH₂), 25.98 (CH), 23.01 (2 × CH₃), 22.44 (2 × CH₃), 20,78 (CH₃-Ar).

Thymyl Hexanoate (3h)

Chromatographic purification gave colorless oil, $C_{16}H_{24}O_2$ (M=248.36); yield 79%; MS (EI): m/z (%) 248 (M⁺) (7.3), 151 (8.9), 150 (79.2), 136 (9.4), 135 (100), 105 (6.7), 99 (6.6), 91 (10.5), 71 (9.7), 43 (9.3); RI (HP5-MS): 1738; 1H NMR (500.13 MHz, CDCl₃): δ (ppm) 7.20 (1H,d, J=8 Hz, Ar-H), 7.03 (1H, d, J=7.6 Hz, Ar-H), 6.81 (1H, s, Ar-H), 2.98 (1H, spt, J=6.9 Hz, CH), 2.58 (2H, t, J=7.6 Hz, CH₂), 2.32 (3H, s, CH₃), 1.79 (2H, quin, J=7.5 Hz, CH₂), 1.41 (4H, m, CH₂), 1.20 (6H, d, J=6.9 Hz, CH₃), 0.95 (3H, m, CH₃); 13 C NMR (125.76 MHz, CDCl₃): δ (ppm) 172.50 (C=O), 147.90 (C_{Ar}), 137.00 (C_{Ar}), 136.50 (C_{Ar}), 127.00 (C_{Ar}), 126.3 (C_{Ar}), 122.7 (C_{Ar}), 34.38 (CH₂), 31.36 (CH₂), 27.08 (CH), 24.76 (CH₂), 23.03 (2 × CH₃), 22.36 (CH₂), 20.83 (CH₃), 13.94 (CH₃).

Thymyl Heptanoate (3i)

Chromatographic purification gave colorless oil. $C_{17}H_{26}O_2$ (M=262.39); yield 84%; MS (EI): m/z (%) 262 (M⁺) (7.6), 151 (10.8), 150 (95.6), 149 (5.3), 136 (9.4), 135 (100), 113 (5.6), 105 (6.4), 91 (9.8), 43 (9.4); RI (HP5-MS): 1840; ¹H NMR (500.13 MHz, CDCl₃): δ (ppm) 7.21 (1H, d, J=8 Hz, Ar-H), 7.03 (1H, d, J=8 Hz, Ar-H), 6.81 (1H, s, Ar-H), 2.99 (1H, spt, J=14 Hz, CH), 2.59 (2H, t, J=8 Hz, CH₂), 2.33 (3H, s, CH₃), 1.79 (2H, m, CH₂), 1.45 (2H, m, CH₂), 1.36 (4H, m, CH₂), 1.22 (6H, m, CH₃), 0.92 (3H, m, CH₃); ¹³C NMR (125.76 MHz, CDCl₃): δ (ppm) 172.53 (C=O), 147.91 (C_{Ar}), 136.99 (C_{Ar}), 136.49 (C_{Ar}), 126.99 (C_{Ar}), 126.33 (C_{Ar}), 122.73 (C_{Ar}), 34.39 (CH₂), 31.44 (CH₂), 28.84 (CH₂), 27.03 (CH), 25.0 (CH₂), 23.0 (2 × CH₃), 22.49 (CH₂), 20.8 (CH₃-Ar), 14.01 (CH₃).

Thymyl Octanoate (3j)

Chromatographic purification gave colorless oil. $C_{18}H_{28}O_2$ (M=276.41); yield 85%; MS (EI): m/z (%) 276 (M⁺) 151 (11.3), 150 (100), 136 (9.3), 135 (98.3), 109 (6.4), 105 (7.2), 91 (10.9), 57 (27.2), 55 (9.4), 43 (6.3); RI (HP5-MS): 1938; 1 H NMR (500.13 MHz, CDCl₃), δ (ppm) 7.21 (1H, d, J=7.9 Hz, Ar-H), 7.03 (1H, d, J=7.9 Hz, Ar-H), 6.81 (1H, s, Ar-H), 2.99 (1H, spt, J=6.9 Hz, CH), 2.59 (2H, t, J=7.5 Hz, CH₂), 2.33 (3H, s, CH₃), 1.79 (2H, quin, J=7.5 Hz, CH₂), 1.47–1.29 (8H, m, CH₂), 1.21 (6H, bd, CH₃), 0.92 (3H, t, J=6.8 Hz, CH₃); 13 C NMR (125.76 MHz, CDCl₃): δ (ppm) 174.20 (C=O), 148.00 (C_{Ar}), 136.80 (C_{Ar}), 136.09 (C_{Ar}), 127.00 (C_{Ar}), 126.30 (C_{Ar}), 122.80 (C_{Ar}), 36.10 (CH₂), 34.00 (CH₂), 32.04 (CH₂), 29.00 (CH₂), 28.00 (CH), 26.00 (CH₂), 23.0 (2 × CH₃), 22.10 (CH₂), 20.5 (CH₃-Ar), 14.00 (CH₃).

Thymyl Nonanoate (3k)

Chromatographic purification gave colorless oil. $C_{19}H_{30}O_2$ (M = 290.44); yield 79.80%; MS (EI): m/z (%)

290 (M⁺) (5.7), 151 (12.0), 150 (100), 141 (3.2), 136 (7.7), 135 (79.0), 121 (3.4), 117 (2.0), 115 (2.3), 91 (3.3); RI (HP5-MS): 2044; ¹H NMR (500.13 MHz, CDCl₃): δ (ppm) 7.24–7.19 (1H, m, Ar-H), 7.06–7.01 (1H, bd, Ar-H), 6.84–6.79 (1H, bs, Ar-H), 3.04–2.93 (1H, bm, CH), 2.63–2.56 (2H, bm, CH₂), 2.36–2.31 (3H, s, CH₃), 1.84–1.75 (2H, bm, CH₂), 1.49–1.26 (10H, bm, CH₂), 1.24–1.18 (6H, bt, CH₃), 0.95–0.87 (3H, bm, CH₃); ¹³C NMR (125.76 MHz, CDCl₃): δ (ppm) 172.53 (C=O), 147.91 (C_{Ar}), 136.99 (C_{Ar}), 136.49 (C_{Ar}), 126.99 (C_{Ar}), 126.33 (C_{Ar}), 122.73 (C_{Ar}), 36.10 (CH₂), 34.20 (CH₂), 32.00 (CH₂), 28.80 (2 × CH₂), 27.60 (CH), 25.60 (CH₂), 23.0 (2 × CH₃), 22.50 (CH₂), 20.8 (CH₃-Ar), 14.01 (CH₃).

Thymyl Decanoate (31)

Chromatographic purification gave colorless oil. $C_{20}H_{32}O_2$ (*M* = 304.47); yield 75.50%; MS (EI): m/z (%): 304 (M⁺) (4.6), 151 (12.3), 150 (100), 136 (6.7), 135 (71.9), 109 (5.3), 91 (5.7), 71 (4.8), 57 (4.8), 55 (5.5); RI (HP5-MS): 2147; ¹H NMR (500.13 MHz, CDCl₃): δ (ppm) 7.21 (1H, d, J = 8 Hz, Ar-H), 7.03 (1H, d, J = 7.6Hz, Ar-H), 6.82 (1H, s, Ar-H), 2.99 (1H, spt, J = 6.9 Hz, CH), 2.59 (2H, t, J = 7.5 Hz, CH₂), 2.33 (3H, s, CH₃), 1.8 (2H, quin, J = 7.5 Hz, CH₂), 1.49-1.25 (12H, bm, CH₂),1.21 (6H, d, J = 6.9 Hz, CH₂), 0.91 (3H, t, J = 6.2 Hz, CH₂); ¹³C NMR (125.76 MHz, CDCl₂): δ (ppm) 172.52 (C=O), 147.91 (C_{Ar}) , 136.99 (C_{Ar}) , 136.48 (C_{Ar}) , 126.99 (C_{Ar}) , 126.33 (C_{Ar}) , 122.73 (C_{Ar}) , 34.39 (CH_2) , 31.85 (CH_2) , 29.43 (CH_2) , 29.25 $(2 \times CH_2)$, 29.17 (CH_2) , 27.04 (CH), 25.05 (CH₂), 23.0 (2 × CH₃), 22.67 (CH₂), 20.8 (CH_3-Ar) , 14.09 (CH_3) .

Thymyl Undecanoate (3m)

Chromatographic purification gave colorless oil. $C_{21}H_{34}O_2$ (*M* = 318.49); yield 82.00%; MS (EI): m/z (%) 318 (M⁺) 151 (7.9), 150 (100), 149 (7.4), 135 (80.1), 134 (6.8), 105 (7.0), 91 (13.3), 57 (18.4), 55 (20.2), 43 (12.1); RI (HP5-MS): 2245; ¹H NMR (500.13 MHz, CDCl₂): δ (ppm) 7.21 (1H, d, J = 8 Hz, Ar-H), 7.03 (1H, d, J = 7.6Hz, Ar-H), 6.82 (1H, s, Ar-H), 2.99 (1H, spt, J = 13.9 Hz, CH), 2.59 (2 H, t, J = 7.46 Hz, CH₂), 2.33 (3H, s, CH₃), 1.79 (2H, quin, J = 7.54 Hz, CH₂), 1.41–1.48 (2H, m, CH_2), 1.26–1.39 (12H, m, CH_2), 1.21 (6H, d, J = 6.94 Hz, CH_3), 0.91 (3H, t, J = 6.76 Hz, CH_3); ¹³C NMR (125.76) MHz, CDCl₃): δ (ppm) 172.52 (C=O), 147.91 (C_{\textit{A}\text{r}}),} $136.99 (C_{Ar}), 136.48 (C_{Ar}), 126.99 (C_{Ar}), 126.33 (C_{Ar}),$ 122.73 (C_{Ar}), 34.39 (CH₂), 31.88 (CH₂), 29.54 (CH₂), 29.47 (CH₂), 29.30 (CH₂), 29.26 (CH₂), 29.17 (CH₂), 27.04 (CH), 25.05 (CH₂), 23.0 (2 × CH₃), 22.67 (CH₂), 20.8 (CH₃-Ar), 14.10 (CH₃).

Thymyl Dodecanoate (3n)

Chromatographic purification gave colorless oil. $C_{22}H_{36}O_2$ (M = 332.52); yield 76.30%; MS (EI): m/z (%) 332 (M⁺) (4.4), 151 (15.2), 150 (100), 136 (6.7), 135 (67.8), 109 (7.3), 91 (4.6), 57 (6.4), 55 (5.6), 43 (3.6); RI

(HP5-MS): 2355; ¹H NMR (500.13 MHz, CDCl₃): δ (ppm) 7.21 (1H, d, J = 8 Hz, Ar-H), 7.03 (1H, d, J = 8 Hz, Ar-H), 6.81 (1H, s, Ar-H), 2.98 (1H, spt, J = 6.9 Hz, CH), 2.59 (2 H, t, J = 7.6 Hz, CH₂), 2.33 (3H, s, CH₃), 1.79 (2H, quin, J = 7.5 Hz, CH₂), 1.48–1.40 (14H, m, CH₂), 1.20 (6H, d, J = 6.9 Hz, CH₃), 0.90 (3H, t, J = 6.8 Hz, CH₃); ¹³C NMR (125.76 MHz, CDCl₃): δ (ppm) 172.55 (C=O), 147.91 (C_{Ar}), 136.99 (C_{Ar}), 136.49 (C_{Ar}), 126.99 (C_{Ar}), 126.33 (C_{Ar}), 122.73 (C_{Ar}), 34.40 (CH₂), 31.90 (CH₂), 29.60 (2 × CH₂), 29.47 (CH₂), 29.33 (CH₂), 29.18 (CH₂), 29.17 (CH₂), 27.04 (CH), 25.06 (CH₂), 23.01 (2 × CH₃), 22.68 (CH₂), 20.81 (CH₃-Ar), 14.10 (CH₃).

Thymyl Tridecanoate (30)

Chromatographic purification gave colorless oil, $C_{23}H_{38}O_2$ (M = 346.55); yield 85.1%; MS (EI): m/z (%) 346 (M⁺) (2.4), 347 (0.6), 197 (0.9), 152 (0.9), 151 (12.9), 150 (100), 137 (0.4), 136 (5.0), 135 (54.2), 119 (0.5); RI (HP5-MS): 2453; ¹H NMR (500.13 MHz, CDCl₂): δ (ppm) 7.21 (1H, d, J = 7.98 Hz, Ar-H), 7.03 (1H, d, J =7.98 Hz, Ar-H), 6.81 (1H, s, Ar-H), 2.98 (1H, spt, J = 13.9Hz, CH), 2.59 (2 H, t, J = 7.63 Hz, CH₂), 2.33 (3H, s, CH_3), 1.79 (2H, quin, J = 7.54 Hz, CH_2), 1.26–1.48 (16H, m, CH₂), 1.20 (6H, d, J = 6.94 Hz, CH₃), 0.90 (3H, t, J =6.76 Hz, CH₂); ¹³C NMR (125.76 MHz, CDCl₂): δ (ppm) 172.5 (C=O), 147.9 (C_{Ar}), 137 (C_{Ar}), 136.5 (C_{Ar}), 127.0 (C_{Ar}) , 126.3 (C_{Ar}) , 122.7 (C_{Ar}) , 34.43 (CH_2) , 31.94 (CH_2) , $29.67 (2 \times CH_2), 29.51 (CH_2), 29.38 (CH_2), 29.3 (CH_2),$ 29.22 (CH₂), 27.08 (CH), 25.09 (CH₂), 23.04 (2 × CH₃), 22.71 (CH₂), 20.83 (CH₃-Ar), 14.13 (CH₃).

Thymyl Tetradecanoate (3p)

Chromatographic purification gave colorless oil. $C_{24}H_{40}O_2$ (M = 360.57); yield 72.05%; MS (EI): m/z (%) 360 (M⁺) (1.9), 152 (0.9), 151 (12.7), 150 (100), 135 (50.7), 134 (4.8), 121 (1.3), 117 (0.7), 115 (1.9), 109(5.1); RI (HP5-MS): 2555; ¹H NMR (500.13 MHz, CDC l_2): δ (ppm) 7.21 (1H, d, J = 8 Hz, Ar-H), 7.03 (1H, d, J = 87.6 Hz, Ar-H), 6.81 (1H, s, Ar-H), 2.99 (1H, spt, J = 6.9Hz, CH), 2.59 (2H, t, J = 7.6 Hz, CH₂), 2.33 (3H, s, CH₃), 1.79 (2H, quin, J = 7.5 Hz, CH₂), 1.49–1.25 (20H, m, CH_2), 1.21 (6H, d, J = 6.9 Hz, CH_2), 0.91 (3H, t, J = 6.9Hz, CH₃); 13 C NMR (125.76 MHz, CDCl₃): δ (ppm) 172.53 (C=O), 147.93 (C_{Ar}), 136.99 (C_{Ar}), 136.48 (C_{Ar}), 126.99 (C_{Ar}), 126.33 (C_{Ar}), 122.73 (C_{Ar}), 34.39 (C_{H_2}), $31.92 \text{ (CH}_2), 29.67 \text{ (CH}_2), 29.65 \text{ (2} \times \text{CH}_2), 29.60 \text{ (CH}_2),$ 29.48 (CH₂), 29.35 (CH₂), 29.27 (CH₂), 29.19 (CH₂), 27.04 (CH), 25.06 (CH₂), 23.00 (2 × CH₃), 22.68 (CH₂), 20.80 (CH₃-Ar), 14.10 (CH₃).

Thymyl Pentadecanoate (3q)

Chromatographic purification gave colorless oil. $C_{25}H_{42}O_2$ (M=374.60); yield 68.90%; MS (EI): m/z (%) 374 (M⁺) (5.3), 151 (23.9), 150 (100), 149 (7.5), 136 (8.3), 135 (76.4), 109 (11.9), 71 (4.4), 57 (7.0), 55 (6.2); RI (HP5-MS): 2669; ¹H NMR (500.13 MHz, CDCl₃): δ

(ppm) 7.21 (1H, d, J = 8 Hz, Ar-H), 7.03 (1H, d, J = 8 Hz, Ar-H), 6.81 (1H, s, Ar-H), 2.99 (1H, spt, J = 6.90 Hz, CH), 2.59 (2H, t, J = 7.6, CH₂), 2.33 (3H, s, CH₃), 1.79 (2H, quin, J = 7.5 Hz, CH₂), 1.48–1.24 (22H, bm, CH₂), 1.21 (6H, d, J = 6.90 Hz, CH₃), 0.9 (3H, t, J = 6.90 Hz, CH₃); ¹³C NMR (125.76 MHz, CDCl₃): δ (ppm) 172.54 (C=O), 147.92 (C_{Ar}), 136.99 (C_{Ar}), 136.49 (C_{Ar}), 126.99 (C_{Ar}), 126.33 (C_{Ar}), 122.73 (C_{Ar}), 34.40 (CH₂), 31.92 (CH₂), 29.67 (2 × CH₂), 29.65 (2 × CH₂), 29.60 (CH₂), 29.48 (CH₂), 29.35 (CH₂), 29.27 (CH₂), 29.18 (CH₂), 27.04 (CH), 25.06 (CH₂), 23.01 (2 × CH₃), 22.68 (CH₂), 20.81 (CH₃-Ar), 14.10 (CH₃).

Thymyl Hexadecanoate (3r)

Chromatographic purification gave colorless oil. $C_{26}H_{44}O_2$ (M = 388.63); yield 95.06%; MS (EI): m/z (%) 388 (M⁺) (5.8), 150 (100), 135 (65.4), 121 (4.3), 108 (3.4), 105 (4.0), 97 (3.3), 69 (5.6), 55 (6.1), 43 (4.3); RI (HP5-MS): 2772; ¹H NMR (500.13 MHz, CDCl₂): δ (ppm) 7.20 (1H, d, J = 8 Hz, Ar-H), 7.03 (1H, d, J = 8 Hz, Ar-H), 6.81 (1H, s, Ar-H), 2.98 (1H, spt, J = 6.90 Hz, CH), 2.58 (2H, t, J = 7.5, CH₂), 2.33 (3H, s, CH₂), 1.78 (2H, quin, J = 7.5 Hz, CH₂), 1.51-1.28 (24H, bm, CH₂),1.21 (6H, d, J = 6.90 Hz, CH₃), 0.9 (3H, t, J = 6.90 Hz, CH₂); 13 C NMR (125.76 MHz, CDCl₂): δ (ppm) 172.50 (C=O), 147.94 (C_{Ar}) , 136.99 (C_{Ar}) , 136.49 (C_{Ar}) , 126.99 (C_{Ar}) , 126.33 (CH_{Ar}) , 122.73 (C_{Ar}) , 34.40 (CH_2) , 31.92 (CH_2) , 29.68 (2 × CH_2), 29.65 (2 × CH_2), 29.60 (CH_2), 29.47 (CH₂), 29.35 (CH₂), 29.27 (CH₂), 29.19 (CH₂), $25.06 \text{ (CH}_2)$, $23.01 \text{ (2} \times \text{CH}_3)$, $22.68 \text{ (CH}_2)$, 20.81 (CH_3 -Ar), 14.10 (CH₂).

Thymyl Heptadecanoate (3s)

Chromatographic purification gave colorless oil. $C_{27}H_{46}O_2$ (*M* = 402.65); yield 72.50%; MS (EI): m/z (%) 402 (M⁺), 151 (19.6), 150 (100), 136 (5.4), 135 (55.2), 109 (9.2), 71 (3.3), 69 (3.0), 57 (5.5), 55 (5.0), 43 (3.0); RI (HP5-MS): 2870; ¹H NMR (500.13 MHz, CDCl₂): δ (ppm) 7.21 (1H, d, J = 7.9 Hz, Ar-H), 7.03 (1H, d, J = 7.9Hz, Ar-H), 6.82 (1H, s, Ar-H), 2.99 (1H, spt, J = 6.80 Hz, CH), 2.59 (2H, t, J = 7.5, CH₂), 2.33 (3H, s, CH₃), 1.8 (2H, quin, J = 7.5 Hz, CH₂), 1.48-1.25 (26H, bm, CH₂),1.21 (6H, d, J = 6.80 Hz, CH₃), 0.91 (3H, t, J = 6.90 Hz, CH₃); 13 C NMR (125.76 MHz, CDCl₃): δ (ppm) 172.50 (C=O), 147.9 (C_{Ar}), 137.0 (C_{Ar}), 136.5 (C_{Ar}), 127.0 (C_{Ar}), 126.30 (CH_{Ar}), 122.70 (C_{Ar}), 34.43 (CH₂), 31.96 (CH₂), $29.72 (2 \times CH_2), 29.52 (2 \times CH_2), 29.40 (CH_2), 29.31$ (CH_2) , 29.22 (CH_2) , 26.54 (CH_2) , 25.09 (CH_2) , 23.05 (2×10^{-2}) CH₃), 22.72 (CH₂), 20.84 (CH₃-Ar), 14.15 (CH₃).

Thymyl Benzoate (3t)

Chromatographic purification gave white solid. $C_{17}H_{18}O_2$ (M = 254.32); yield 95%; MS (EI): m/z (%): 254 (M⁺), (9.7), 150 (2.7), 149 (25.0), 133 (2.3), 106 (9.4), 105 (100), 91 (5.1), 78 (3.1), 77 (34.5), 51 (4.1); RI (HP5-MS): 1955; ¹H NMR (500.13 MHz, CDCl₃): δ (ppm) 8.26

(2H, d, J = 7.3 Hz, Ar-H), 7.68 (1H, t, Ar-H), 7.56 (2H, t, Ar-H), 7.28 (1H, d, J = 8 Hz, Ar-H), 7.11 (1H, d, J = 8 Hz, Ar-H), 6.98 (1H, s, Ar-H), 3.10 (1H, spt, J = 6.90 Hz, CH), 2.38 (3H, s, CH₃), 1.25 (6H, d, J = 6.90 Hz, CH₃); 13 C NMR (125.76 MHz, CDCl₃): δ (ppm) 165.33 (C=O), 148.11 (C_{Ar}), 137.16 (C_{Ar}), 136.62 (C_{Ar}), 133.50 (C_{Ar}), 130.12 (2 × CH_{Ar}), 129.63 (CH_{Ar}), 128.60 (2 × CH_{Ar}), 127.15 (CH_{Ar}), 126.45 (CH_{Ar}), 122.84 (CH_{Ar}), 27.24 (CH), 23.03 (2 × CH₃), 20.85 (CH₃-Ar).

2. 3. Identification of Synthetized Compounds

2. 3. 1. GC-MS Analysis

MS spectra of samples of the synthesized compounds were recorded on a 7890/7000B GC/MS/MS triple quadrupole system (Agilent Technologies, USA, equipped with a Combi PAL auto sampler). The fused silica capillary column HP-5MS (5% phenylmethylsiloxane, $30 \text{ m} \times 0.25 \text{ mm}$, film thickness 0.25 im, Agilent Technologies, Palo Alto, CA, USA) was used. The injector, source and interface operated at 250, 230 and 300 °C, respectively. The temperature program: from 60 for 5 min isothermal to 300 °C at a heating rate of 8 °C/min and on 300 °C for 5 min isothermal. The solutions in hexane were injected in split ratio 10:1. The carrier gas was helium with a flow of 1.0 mL/min. Post run: back flash for 1.89 min, at 280 °C, with helium at 50 psi. MS conditions were as follows: ionization voltage of 70 eV, acquisition mass range 50-650, scan time 0.32 s. Semi-quantitative analysis was carried out directly from peak areas in the GC profile.

Linear retention indices (RI) were determined based on the retention times of $\rm C_8$ – $\rm C_{40}$ alkanes run on HP-5MS column using the above mentioned temperature programme. 41

2. 3. 2. NMR Analysis

¹H NMR spectra were recorded in CDCl₃ (isotopic enrichment 99.95%) solutions at 25 °C using a Bruker AVANCE 500 instrument (500.13 MHz for ¹H, 125.76 MHz for ¹³C) using 5 mm inverse detection broadband probes and deuterium lock.

2. 4. Antimicrobial Activity

2. 4. 1. Microbial Strains

The *in vitro* antimicrobial activity of the synthesized compounds was tested against a panel of laboratory control strains belonging to the American Type Culture Collection Maryland, USA. Gram-positive: *Bacillus subtilis* ATCC 6633 and *Staphylococcus aureus* ATCC 6538; Gram-negative: *Escherichia coli* ATCC 8739, *Pseudomonas aeruginosa* ATCC 9027; fungal organisms: *Aspergil*-

lus niger ATCC 16404 and Candida albicans ATCC 10231. The Gram-negative bacteria Salmonella abony NCTC 6017 and Salmonela typhimurium ATCC 14028 were obtained from the National Collection of Type Cultures. All microorganisms were maintained at –20 °C under appropriate conditions and regenerated twice before use in the manipulations.

2. 4. 2. Screening of Antimicrobial Activity

The minimal inhibitory concentration (MIC) of esters was determined based on a broth microdilution method in 96-well microtitre plates. ⁴² The inocula of the bacterial strains were prepared from overnight broth cultures and suspensions were adjusted to 0.5 McFarland standard turbidity. Dimethyl sulphoxide (10% aqueous solution) was used to dissolve and to dilute samples to the highest concentration to be tested (stock concentrations 1 mg/m-L). A serial doubling dilution of the samples was prepared in a 96-well microtiter plate, using the method of Sarker et al. ⁴³ with slight modifications. The minimal bactericidal/fungicidal concentration (MBC/MFC) was evaluated as the lowest concentration of tested samples at which inoculated microorganisms were 99.9% killed. Tests were carried out in triplicate.

2. 5. *In silico* Physico-chemical, Pharmacokinetic and Toxicological Properties of the Synthetized Compounds

In order to obtain a complete picture of the synthesized compounds **3a–t** an *in silico* study was performed. Physico-chemical, pharmacokinetic and toxicological properties of compounds were calculated using the Molinspiration,⁴⁴ admetSAR,⁴⁵ DataWarrior⁴⁶ and Toxtree prediction tools.⁴⁷

3. Results and Discussion

3. 1. Chemical Synthesis

A small focused library of twenty thymyl esters was synthesized. To the best of our knowledge ten of twenty compounds are new (i.e. 3i, 3k-s; Scheme 1).

Although enzyme-catalyzed esterification of alcohols of different structures is a well-established approach, 48 the enzymatic esterification of phenols is not frequently reported. 49 Since a biocatalytic approach appeared especially appealing to us, we have tried to repeat the experimental protocol reported for the esterification of functionalized phenols with *Candida antarctica* lipase (CAL-B)49 using thymol and propanoic acid as substrates and *tert*-butyl methyl ether and hexane as solvents. However, although no reaction occurred in our hands, the biocatalytic approach deserves further investigation within the recent trends of green chemistry methodologies.

Scheme 2. Synthesis of thymyl esters: (a) chemical synthesis approach; (b) enzyme-catalyzed approach.

Table 1. Thymyl esters: chemical entity, yields (%) and entry

R	Yield	Product Name and Number	•
	(%)		
CH ₃	85.00	Thymyl Acetate	3a
CH ₂ Cl	85.00	Thymyl 2-Chloroacetate	3b
CH ₂ CH ₃	82.00	Thymyl Propanoate	3c
CH ₂ CH ₂ CH ₃	84.00	Thymyl Butanoate	3d
CH(CH ₃) ₂	75.00	Thymyl 2-Methylpropanoate	3e
CH ₂ (CH ₂) ₂ CH ₃	87.00	Thymyl Pentanoate	3f
$CH_2CH(CH_3)_2$	81.00	Thymyl 3-Methylbutanoate	3g
$CH_2(CH_2)_3CH_3$	79.00	Thymyl Hexanoate	3h
$CH_2(CH_2)_4CH_3$	84.00	Thymyl Heptanoate	3i
CH ₂ (CH ₂) ₅ CH ₃	85.00	Thymyl Octanoate	3j
CH ₂ (CH ₂) ₆ CH ₃	79.80	Thymyl Nonanoate	3k
$CH_2(CH_2)_7CH_3$	75.50	Thymyl Decanoate	31
CH ₂ (CH ₂) ₈ CH ₃	82.00	Thymyl Undecanoate	3m
$CH_2(CH_2)_9CH_3$	76.30	Thymyl Dodecanoate	3n
$CH_2(CH_2)_{10}CH_3$	85.10	Thymyl Tridecanoate	30
$CH_{2}(CH_{2})_{11}CH_{3}$	72.05	Thymyl Tetradecanoate	3p
$CH_{2}(CH_{2})_{12}CH_{3}$	68.90	Thymyl Pentadecanoate	3q
$CH_{2}(CH_{2})_{13}CH_{3}$	95.06	Thymyl Hexadecanoate	3r
$CH_{2}(CH_{2})_{14}CH_{3}$	72.50	Thymyl Heptadecanoate	3s
Ph	95.00	Thymyl Benzoate	3t

3. 2. Antimicrobial Activity

The results obtained in broth microdilution assay are presented in Supplementary data, Table S1. The assayed

samples were less effective than antibiotic/antimycotic used as reference standard and if noted, activity was never greater than the values obtained for the parent compound 1 (MIC/MBC/MFC never exceeded 0.5 mg/mL, Supplementary data, Table S1). The results are indicating selective susceptibility of the microorganisms, with *S. aureus* (3a,b,e), *P. aeruginosa* (3b,j,k,p) and *C. albicans* (3a–e,g,n,p) being the most sensitive strains to synthesized derivatives. On the other hand, five microorganisms (*B. subtilis, E. coli, S. abony, S. typhimurium* and *A. niger*) were completely resistant to synthesized compounds tested (initial concentration 1 mg/mL).

Five of our samples (3a,c,e,g,t) are matching the samples tested by Mathela and collaborators, 4 who were making evaluation of antibacterial activity on Streptococcus mutans (MTCC 890), S. aureus (MTCC 96), B. subtilis (MTCC 121), Staphylococcus epidermidis (MTCC 435) and E. coli (MTCC 723), and who reported the enhancement of the activity for esters in comparison to thymol. For all other synthesized compounds antimicrobial results are reported for the first time. An interesting fact is that MIC values by Mathela⁴ were three- (2-methylpropanoate 3c and 3-methylbutanoate 3e) to even ten- (acetate **3a**) times lower for *B. subtilis* than for thymol. Having in best case comparable, but never greater MIC values than for thymol itself, in our in vitro experiment we could not confirm such results.⁴ The importance of free hydroxyl group in the phenolic structure was confirmed in terms of activity when carvacrol was compared to its methyl ether,⁵⁰ however results presented by Mathela⁴ are contrary to the above-mentioned fact and to our results (Supplementary data, Table S1).

3. 3. In silico Study

3. 3. 1. Physico-chemical Properties of the Thymyl Esters 3a-t

Lipinski's rule of 5 gives evaluation to drug-likeness and determines if a substance with certain pharmacological or biological activity has properties that would make it a likely orally active drug in humans. ⁵¹ Calculations of important molecular parametars stated by Lipinski such as fragment based contributions and correction factors (m_iLog P), topological polar surface area (TPSA) and drug-likeness were obtained by Molinspiration software. ⁴⁴ Drug-likeness score of compounds **3a–t** is given in Suplementary data, Table S2.

Seven compounds ($3\mathbf{a}-\mathbf{g}$) had $\mathbf{m_i}\mathrm{LogP}$ values below 5 and thus were predicted to have sufficient oral bioavailability. The rest of the compounds were estimated as lypophilic. All of the tested compounds ($3\mathbf{a}-\mathbf{t}$) had TPSA below 60 Ų, and thus were predicted to have both good intestinal absorption and good BBB penetration. All of the tested compounds ($3\mathbf{a}-\mathbf{t}$) had less than 10 H-bond acceptors (\mathbf{n}_{ON}) and less than 5 H-bond donors (\mathbf{n}_{OHNH}). The conformational flexibility, described by the number of rotatable bonds (\mathbf{n}_{rotb}), for 12 compounds ($3\mathbf{a}-\mathbf{k},\mathbf{t}$) was between 0 and 10, which warrants good oral bioavailability. Finally, all compounds ($3\mathbf{a}-\mathbf{t}$) had both molecular weight (MW) and molecular volume below 500.

Calculated physico-chemical properties showed that seven thymyl esters (3a–g) were predicted to have good oral bioavailability, with values for fragment based contributions and correction factors $m_iLogP < 5$, TPSA < 140, MW < 500, $n_{ON} < 10$, $n_{OHNH} < 5$ and $n_{rotb} < 10$.

3. 3. 2. Pharmacokinetic Properties of the Thymyl Esters 3a-t

Absorption properties of compounds **3a–t** were predicted by admetSAR⁴⁵ (Supplementary data, Table S3). The results suggested that all of the tested compounds (**3a–t**) might be able to pass through blood-brain barrier (BBB) and penetrate into the CNS, might be capable of being absorbed by intestine, and were supposed to have positive Caco-2 permeability. Moreover, compounds **3a–t** were predicted as non-substrates for P-glycoprotein, non-inhibitors of P-glycoprotein, and as non-inhibitors against renal organic cation transporter (ROCT).

Metabolic properties of the thymyl esters **3a–t** were predicted by admetSAR⁴⁵ (Supplementary data, Table S4). None of the compounds was predicted as CYP450 2C9 and 2D6 substrate, while 16 compounds (**3b,d,f,h–t**) were predicted as CYP450 3A4 substrates. All of the te-

sted compounds were predicted as CYP450 1A2 inhibitors, but none of them was predicted as CYP450 2D6 and 3A4 inhibitors. Moreover, 15 compounds (3b,d,h-t) might be able to inhibit CYP450 2C19 enzyme, while only compound 3t was predicted as CYP450 2C9 inhibitor. Almost all thymyl esters (3a-s) were predicted to have low CYP inhibitory promiscuity, except compound 3t.

3. 3. 3. Toxicological Properties of the Thymyl Esters 3a-t

The structural alerts for DNA and protein binding for compounds $\bf 3a-t$ were predicted using Toxtree prediction tool based on decision tree approach. Tompounds $\bf 3a-t$ showed structural alerts for DNA binding as they were predicted as compounds able to undergo Michael addition. Moreover, compounds $\bf 3a-t$ showed structural alerts for protein binding due to their predicted ability to undergo Michael addition, ability to participate in acyl transfer and ability to undergo $\bf S_N 2$ reactions (results given in Supplementary data, Table S5).

Toxicological properties of compounds **3a–t** predicted by admetSAR⁴⁵ have characterized compounds **3a–t** as weak HERG (human Ether-à-go-go-Related Gene) inhibitors, non-AMES toxic and non-carcinogens, but highly toxic for fish, *Tetrahymena pyriformis* and honey bee. Ready biodegradable were supposed to be six compounds (**3a,c–g**). Depending on the risk for acute oral toxicity, only compound **3b** was predicted as Category II, or compound with LD₅₀ value greater than 50 mg/kg but less than 500 mg/kg, while the rest of the tested compounds (**3a,c–t**) were predicted as Category III, or compounds with LD₅₀ values greater than 500 mg/kg but less than 5000 mg/kg. According to the TD₅₀ values, compounds **3a–t** were predicted as »non-required« or non-carcinogenic chemicals (Supplementary data, Table S6).

Only one study involving thymyl esters was undertaken to establish the potential acute toxicity in animal models.²⁹ The esters shown no acute toxicity to mice at doses higher than 5000 mg/kg which represents good congruence and indicate the usefulness of data obtained in our *in silico* study.

Toxicological properties of thymyl esters **3a–t** predicted by DataWarrior⁴⁶ have shown that only compound **3b** has a high risk for all mutagenic, tumorigenic, reproductive and irritant effects. Compound **3h** was predicted to have high risk for tumorigenic and irritant effects. However, all compounds were predicted to have high risk for irritant effects (Supplementary data, Table S7).

4. Conclusion

We synthesized twenty esters of thymol, of which ten represent new compounds. All of the compounds were employed in antimicrobial bioassay and was found that lower representatives of the synthesized homologous series of esters are antimicrobials comparable to thymol and can be considered as activity key holders, too. Results of our in silico study predicted that seven esters (lower representatives and short-chain fatty acids esters 3a-g) obey Lipinski's rule of five, showing drug-likeness. The rest of the compounds were estimated as lipophilic. All compounds, except thymyl 2-chloroacetate (3b) and thymyl hexanoate (3h), were predicted as non-mutagenic, non-tumorigenic, non-AMES toxic and non-carcinogenic, but highly toxic for fish, T. pyriformis and honey bee. They are likely to be absorbed by intestine and were predicted as ready biodegradable, weak HERG inhibitors, Category III of risk for acute toxicity, with no risk for reproductive effects, but with high risk for irritant effects. Taking in consideration predicted in silico properties and estimated drug likeness score, pharmacological and toxicological profile, thymyl esters might be used as prodrugs. Among the chemical bonds used to link parental drug and carrier, esters have already proven to be promising due to their amenability to hydrolysis in vivo and are most frequently used in order to enhance the lipophilicity⁵² and passive membrane transport.

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Povzetek

Z derivatizacijo izhodnih struktur terpenoidov lahko pogosto pripravimo nove spojine, ki imajo izboljšane biološke aktivnosti. Naravni fenolov derivat timol, ki biosintezno nastane po terpenski poti, predstavlja dobro znan biocid z močnim antimikrobnim delovanjem in različnimi drugimi terapevtskimi aktivnostmi. Namen naše študije je bil pripraviti majhno, fokusirano knjižnico dvajsetih timilnih estrov s pomočjo ene same modifikacije fenolne funkcionalne skupine v timolu. Deset izmed pripravljenih spojin je novih. Vsem sintetiziranim spojinam smo *in vitro* določili antimikrobne lastnosti. Drug pomemben aspekt naše študije pa je bila uporaba *in silico* računskih metod za določitev fizikalno-kemijskih, farmakokinetičnih in toksikoloških lastnosti spojin, kar je omogočilo dodaten vpogled v njihove aktivnosti in daje nove usmeritve za nadaljnje raziskave.